

NaO Structure:

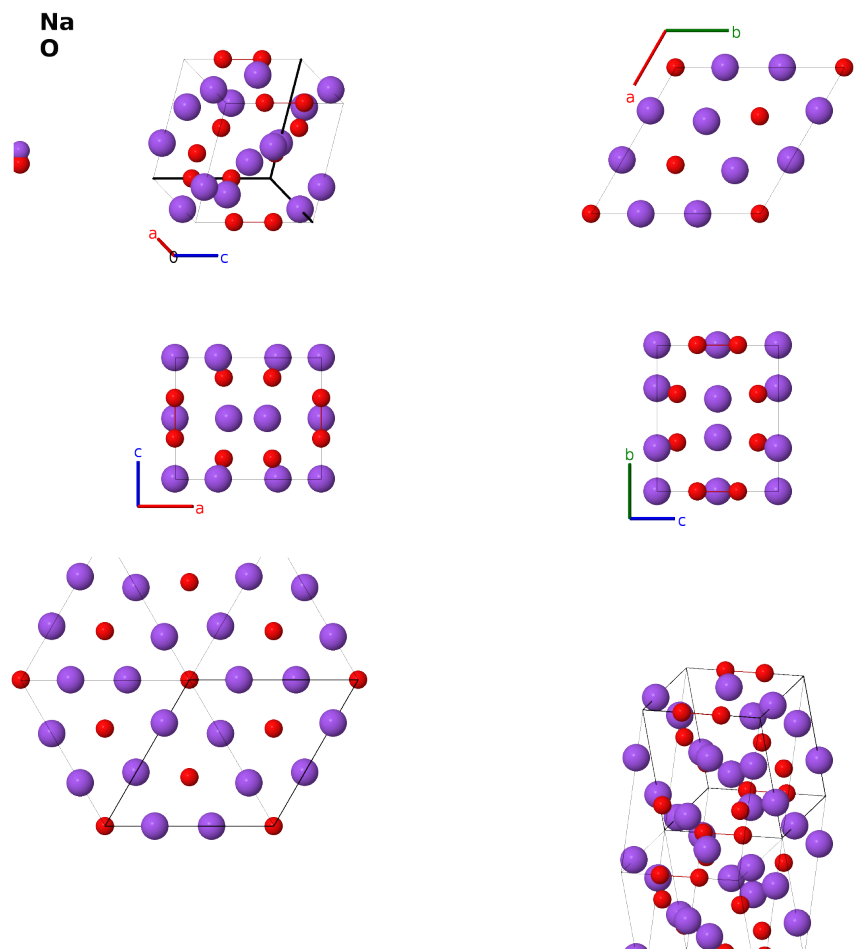
AB_hP12_189_fg_eh-001

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- D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/prototype-encyclopedia/PQWK>

https://aflow.org/prototype-encyclopedia/AB_hP12_189_fg_eh-001



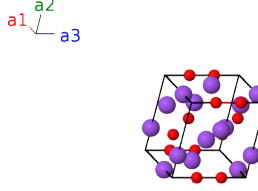
Prototype	NaO
AFLOW prototype label	AB_hP12_189_fg_eh-001
ICSD	25526
CCDC	1601180
Pearson symbol	hP12
Space group number	189
Space group symbol	$P\bar{6}2m$

AFLOW prototype command `aflow --proto=AB_hP12_189_fg_eh-001`
 `--params=a, c/a, z1, x2, x3, z4`

Other compounds with this structure

CaAs, CaP, EuAs, KS, KSe, α -KTe, α -NaS, β -RbS, RbSe, RbTc, SrAs, SrP

Hexagonal primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}} \end{aligned}$$


Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= z_1 \mathbf{a}_3$	$=$	$c z_1 \hat{\mathbf{z}}$	(2e)	O I
$\mathbf{B}_2$	$= -z_1 \mathbf{a}_3$	$=$	$-c z_1 \hat{\mathbf{z}}$	(2e)	O I
$\mathbf{B}_3$	$= x_2 \mathbf{a}_1$	$=$	$\frac{1}{2}a x_2 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a x_2 \hat{\mathbf{y}}$	(3f)	Na I
$\mathbf{B}_4$	$= x_2 \mathbf{a}_2$	$=$	$\frac{1}{2}a x_2 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a x_2 \hat{\mathbf{y}}$	(3f)	Na I
$\mathbf{B}_5$	$= -x_2 \mathbf{a}_1 - x_2 \mathbf{a}_2$	$=$	$-a x_2 \hat{\mathbf{x}}$	(3f)	Na I
$\mathbf{B}_6$	$= x_3 \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a x_3 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a x_3 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3g)	Na II
$\mathbf{B}_7$	$= x_3 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a x_3 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a x_3 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3g)	Na II
$\mathbf{B}_8$	$= -x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$-a x_3 \hat{\mathbf{x}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3g)	Na II
$\mathbf{B}_9$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_4 \hat{\mathbf{z}}$	(4h)	O II
$\mathbf{B}_{10}$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - c z_4 \hat{\mathbf{z}}$	(4h)	O II
$\mathbf{B}_{11}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - c z_4 \hat{\mathbf{z}}$	(4h)	O II
$\mathbf{B}_{12}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_4 \hat{\mathbf{z}}$	(4h)	O II

References

- [1] H. Föppl, *Die Kristallstrukturen der Alkaliperoxyde*, Z. Anorganische und Allgemeine Chemie **291**, 12–50 (1957), doi:10.1002/zaac.19572910104.

First cited in

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