

FeOCl ($E0_5$) Structure: ABC_oP6_59_a_b_a-001

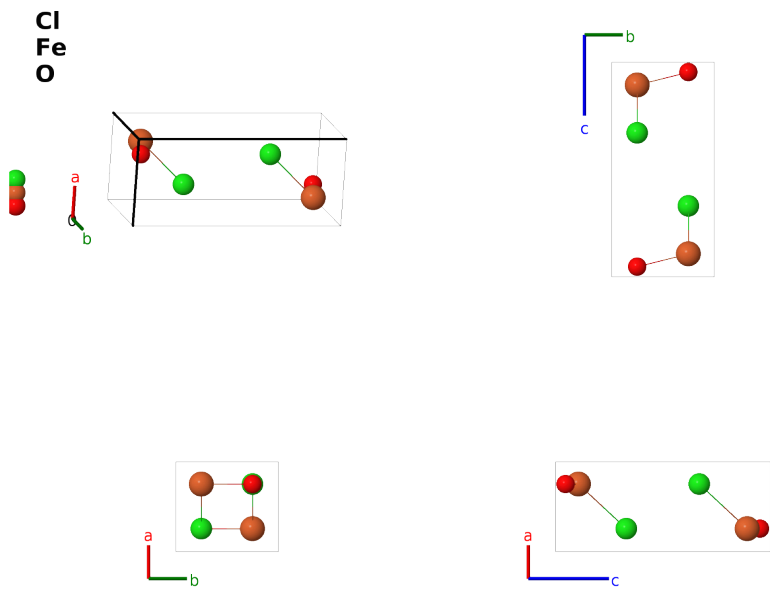
This structure previously used the label(s) **ABC_oP6_59_a_b_a**. Calls to these addresses will be redirected here.

Cite this page as:

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<https://aflow.org/prototype-encyclopedia/DN69>

https://aflow.org/prototype-encyclopedia/ABC_oP6_59_a_b_a-001



Prototype	ClFeO
AFLOW prototype label	ABC_oP6_59_a_b_a-001
<i>Strukturbericht</i> designation	$E0_5$
ICSD	40963
CCDC	1610772
Pearson symbol	oP6
Space group number	59
Space group symbol	$Pm\bar{m}n$
AFLOW prototype command	<code>aflow --proto=ABC_oP6_59_a_b_a-001</code> <code>--params=a,b/a,c/a,z₁,z₂,z₃</code>

Other compounds with this structure

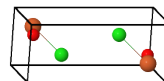
AlOCl, CrOCl, CrOF, CrSBr, HfNBr, HfNCl, HfNI, IOBr, IOCl, TiNBr, TiNCl, TiNI, TiOCl, VOCl, ZrNBr, ZrNCl, ZrNI

Simple Orthorhombic primitive vectors

$$\mathbf{a}_1 = a \hat{\mathbf{x}}$$

$$\mathbf{a}_2 = b \hat{\mathbf{y}}$$

$$\mathbf{a}_3 = c \hat{\mathbf{z}}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_1 \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} + cz_1 \hat{\mathbf{z}}$	(2a)	Cl I
$\mathbf{B}_2$	$= \frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_1 \mathbf{a}_3$	$=$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} - cz_1 \hat{\mathbf{z}}$	(2a)	Cl I
$\mathbf{B}_3$	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_2 \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} + cz_2 \hat{\mathbf{z}}$	(2a)	O I
$\mathbf{B}_4$	$= \frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_2 \mathbf{a}_3$	$=$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} - cz_2 \hat{\mathbf{z}}$	(2a)	O I
$\mathbf{B}_5$	$= \frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(2b)	Fe I
$\mathbf{B}_6$	$= \frac{3}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(2b)	Fe I

References

- [1] S. M. Kauzlarich, J. L. Stanton, J. Faber, and B. A. Averill, *Neutron profile refinement of the structure of FeOCl and FeOCl(TTF)1/8.5*, J. Am. Chem. Soc. **108**, 7946–7951 (1986), doi:10.1021/ja00285a011.

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