

Sodium Chlorate (NaClO_3 , $G0_3$) Structure: ABC3_cP20_198_a_a_b-001

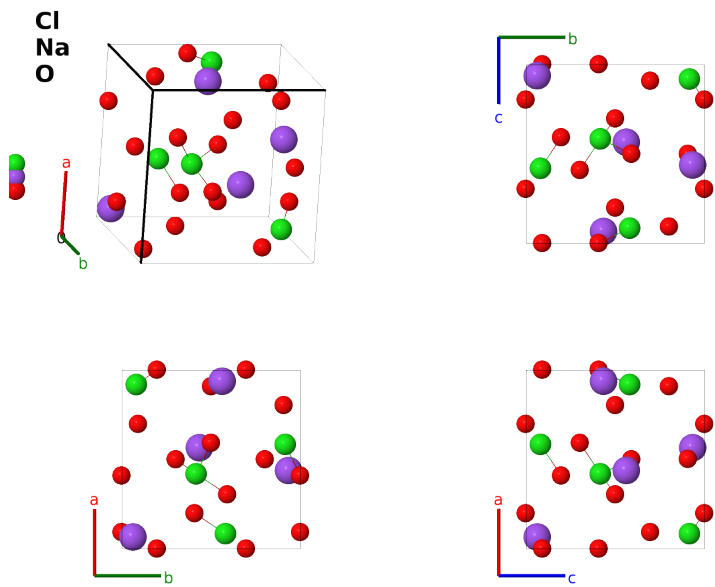
This structure previously used the label(s) **ABC3_cP20_198_a_a_b**. Calls to these addresses will be redirected here.

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<https://aflow.org/prototype-encyclopedia/QEXW>

https://aflow.org/prototype-encyclopedia/ABC3_cP20_198_a_a_b-001



Prototype	NaClO_3
AFLOW prototype label	ABC3_cP20_198_a_a_b-001
<i>Strukturbericht</i> designation	$G0_3$
Mineral name	sodium chlorate
ICSD	16714
CCDC	1597266
Pearson symbol	cP20
Space group number	198
Space group symbol	$P2_13$
AFLOW prototype command	<code>aflow --proto=ABC3_cP20_198_a_a_b-001 --params=a, x1, x2, x3, y3, z3</code>

Other compounds with this structure

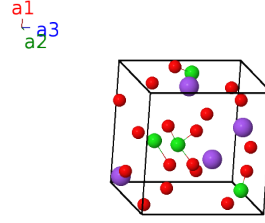
NaBrO₃

Simple Cubic primitive vectors

$$\mathbf{a}_1 = a \hat{\mathbf{x}}$$

$$\mathbf{a}_2 = a \hat{\mathbf{y}}$$

$$\mathbf{a}_3 = a \hat{\mathbf{z}}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= x_1 \mathbf{a}_1 + x_1 \mathbf{a}_2 + x_1 \mathbf{a}_3$	$=$	$a x_1 \hat{\mathbf{x}} + a x_1 \hat{\mathbf{y}} + a x_1 \hat{\mathbf{z}}$	(4a)	Cl I
$\mathbf{B}_2$	$= -\left(x_1 - \frac{1}{2}\right) \mathbf{a}_1 - x_1 \mathbf{a}_2 + \left(x_1 + \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a \left(x_1 - \frac{1}{2}\right) \hat{\mathbf{x}} - a x_1 \hat{\mathbf{y}} + a \left(x_1 + \frac{1}{2}\right) \hat{\mathbf{z}}$	(4a)	Cl I
$\mathbf{B}_3$	$= -x_1 \mathbf{a}_1 + \left(x_1 + \frac{1}{2}\right) \mathbf{a}_2 - \left(x_1 - \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a x_1 \hat{\mathbf{x}} + a \left(x_1 + \frac{1}{2}\right) \hat{\mathbf{y}} - a \left(x_1 - \frac{1}{2}\right) \hat{\mathbf{z}}$	(4a)	Cl I
$\mathbf{B}_4$	$= \left(x_1 + \frac{1}{2}\right) \mathbf{a}_1 - \left(x_1 - \frac{1}{2}\right) \mathbf{a}_2 - x_1 \mathbf{a}_3$	$=$	$a \left(x_1 + \frac{1}{2}\right) \hat{\mathbf{x}} - a \left(x_1 - \frac{1}{2}\right) \hat{\mathbf{y}} - a x_1 \hat{\mathbf{z}}$	(4a)	Cl I
$\mathbf{B}_5$	$= x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 + x_2 \mathbf{a}_3$	$=$	$a x_2 \hat{\mathbf{x}} + a x_2 \hat{\mathbf{y}} + a x_2 \hat{\mathbf{z}}$	(4a)	Na I
$\mathbf{B}_6$	$= -\left(x_2 - \frac{1}{2}\right) \mathbf{a}_1 - x_2 \mathbf{a}_2 + \left(x_2 + \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a \left(x_2 - \frac{1}{2}\right) \hat{\mathbf{x}} - a x_2 \hat{\mathbf{y}} + a \left(x_2 + \frac{1}{2}\right) \hat{\mathbf{z}}$	(4a)	Na I
$\mathbf{B}_7$	$= -x_2 \mathbf{a}_1 + \left(x_2 + \frac{1}{2}\right) \mathbf{a}_2 - \left(x_2 - \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a x_2 \hat{\mathbf{x}} + a \left(x_2 + \frac{1}{2}\right) \hat{\mathbf{y}} - a \left(x_2 - \frac{1}{2}\right) \hat{\mathbf{z}}$	(4a)	Na I
$\mathbf{B}_8$	$= \left(x_2 + \frac{1}{2}\right) \mathbf{a}_1 - \left(x_2 - \frac{1}{2}\right) \mathbf{a}_2 - x_2 \mathbf{a}_3$	$=$	$a \left(x_2 + \frac{1}{2}\right) \hat{\mathbf{x}} - a \left(x_2 - \frac{1}{2}\right) \hat{\mathbf{y}} - a x_2 \hat{\mathbf{z}}$	(4a)	Na I
$\mathbf{B}_9$	$= x_3 \mathbf{a}_1 + y_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$a x_3 \hat{\mathbf{x}} + a y_3 \hat{\mathbf{y}} + a z_3 \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{10}$	$= -\left(x_3 - \frac{1}{2}\right) \mathbf{a}_1 - y_3 \mathbf{a}_2 + \left(z_3 + \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a \left(x_3 - \frac{1}{2}\right) \hat{\mathbf{x}} - a y_3 \hat{\mathbf{y}} + a \left(z_3 + \frac{1}{2}\right) \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{11}$	$= -x_3 \mathbf{a}_1 + \left(y_3 + \frac{1}{2}\right) \mathbf{a}_2 - \left(z_3 - \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a x_3 \hat{\mathbf{x}} + a \left(y_3 + \frac{1}{2}\right) \hat{\mathbf{y}} - a \left(z_3 - \frac{1}{2}\right) \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{12}$	$= \left(x_3 + \frac{1}{2}\right) \mathbf{a}_1 - \left(y_3 - \frac{1}{2}\right) \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$a \left(x_3 + \frac{1}{2}\right) \hat{\mathbf{x}} - a \left(y_3 - \frac{1}{2}\right) \hat{\mathbf{y}} - a z_3 \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{13}$	$= z_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + y_3 \mathbf{a}_3$	$=$	$a z_3 \hat{\mathbf{x}} + a x_3 \hat{\mathbf{y}} + a y_3 \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{14}$	$= \left(z_3 + \frac{1}{2}\right) \mathbf{a}_1 - \left(x_3 - \frac{1}{2}\right) \mathbf{a}_2 - y_3 \mathbf{a}_3$	$=$	$a \left(z_3 + \frac{1}{2}\right) \hat{\mathbf{x}} - a \left(x_3 - \frac{1}{2}\right) \hat{\mathbf{y}} - a y_3 \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{15}$	$= -\left(z_3 - \frac{1}{2}\right) \mathbf{a}_1 - x_3 \mathbf{a}_2 + \left(y_3 + \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a \left(z_3 - \frac{1}{2}\right) \hat{\mathbf{x}} - a x_3 \hat{\mathbf{y}} + a \left(y_3 + \frac{1}{2}\right) \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{16}$	$= -z_3 \mathbf{a}_1 + \left(x_3 + \frac{1}{2}\right) \mathbf{a}_2 - \left(y_3 - \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a z_3 \hat{\mathbf{x}} + a \left(x_3 + \frac{1}{2}\right) \hat{\mathbf{y}} - a \left(y_3 - \frac{1}{2}\right) \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{17}$	$= y_3 \mathbf{a}_1 + z_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	$=$	$a y_3 \hat{\mathbf{x}} + a z_3 \hat{\mathbf{y}} + a x_3 \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{18}$	$= -y_3 \mathbf{a}_1 + \left(z_3 + \frac{1}{2}\right) \mathbf{a}_2 - \left(x_3 - \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-a y_3 \hat{\mathbf{x}} + a \left(z_3 + \frac{1}{2}\right) \hat{\mathbf{y}} - a \left(x_3 - \frac{1}{2}\right) \hat{\mathbf{z}}$	(12b)	O I
$\mathbf{B}_{19}$	$= \left(y_3 + \frac{1}{2}\right) \mathbf{a}_1 - \left(z_3 - \frac{1}{2}\right) \mathbf{a}_2 - x_3 \mathbf{a}_3$	$=$	$a \left(y_3 + \frac{1}{2}\right) \hat{\mathbf{x}} - a \left(z_3 - \frac{1}{2}\right) \hat{\mathbf{y}} - a x_3 \hat{\mathbf{z}}$	(12b)	O I

$$\mathbf{B}_{20} = -\left(y_3 - \frac{1}{2}\right) \mathbf{a}_1 - z_3 \mathbf{a}_2 + \left(x_3 + \frac{1}{2}\right) \mathbf{a}_3 = -a\left(y_3 - \frac{1}{2}\right) \hat{\mathbf{x}} - az_3 \hat{\mathbf{y}} + a\left(x_3 + \frac{1}{2}\right) \hat{\mathbf{z}} \quad (12b) \quad \text{O I}$$

References

- [1] G. N. Ramachandran and K. S. Chandrasekaran, *The absolute configuration of sodium chlorate* **10**, 671–675 (1957), doi:10.1107/S0365110X57002327.

Found in

- [1] A. A. Kaminskii, S. N. Bagayev, J. Hulliger, H. Eichler, J. Findeisen, and R. Macdonald, *Acentric cubic NaClO₃—a new crystal for Raman lasers*, Applied Physics B **67**, 157–162 (1998), doi:10.1007/s003400050487.

First cited in

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