

Hexagonal Delafossite (CuFeO₂) Structure: ABC2_hP8_194_a_c-f-005

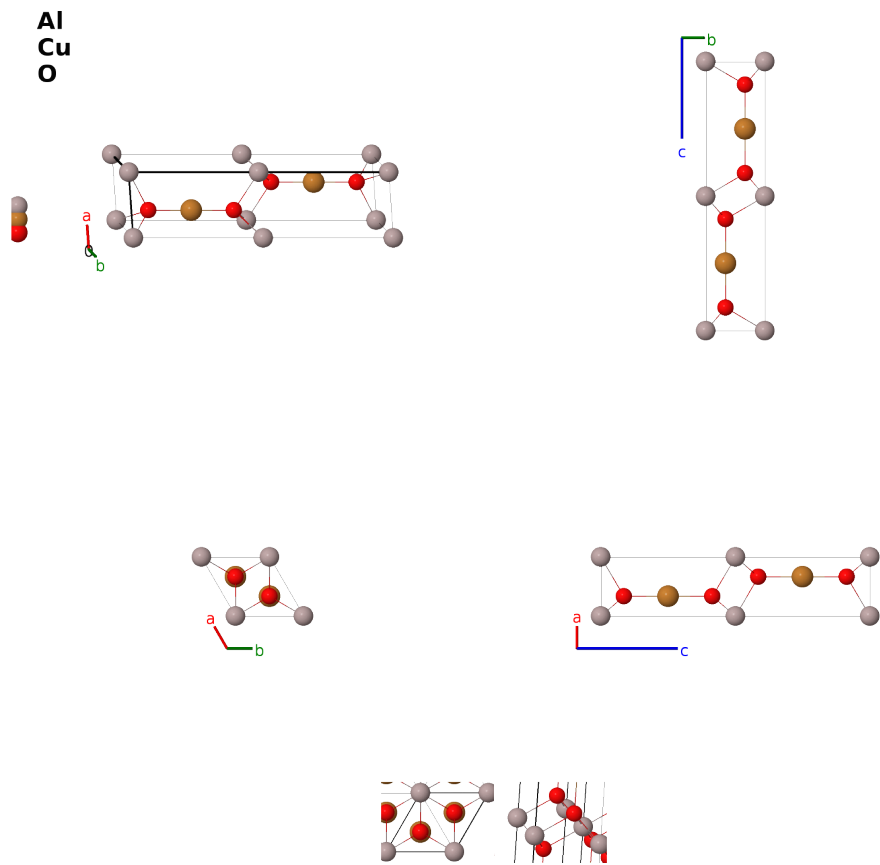
This structure previously used the label(s) **ABC2_hP8_194_a_c-f**. Calls to these addresses will be redirected here.

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<https://aflow.org/prototype-encyclopedia/Y6AL>

https://aflow.org/prototype-encyclopedia/ABC2_hP8_194_a_c-f-005



Prototype	CuFeO ₂
AFLOW prototype label	ABC2_hP8_194_a_c-f-005
Mineral name	delafossite
ICSD	60845
CCDC	1624408
Pearson symbol	hP8
Space group number	194
Space group symbol	$P6_3/mmc$

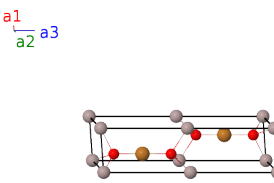
AFLOW prototype command `aflow --proto=ABC2_hP8_194_a_c_f-005`
 `--params=a, c/a, z3`

Other compounds with this structure

AlCCr₂, AlCNb₂, AlCTa₂, AlCTi₂, AlCV₂, AsCNb₂, AsCV₂, AlCuO₂, BBiHf₂, BPbHf₂, CdCTi₂, CuAlO₂, CuGaO₂, CuScO₂, CuYO₂, GaCCr₂, GaCMo₂, GaCNb₂, GaCTa₂, GaCTi₂, GaCV₂, GaNTi₂, GeCCr₂, GeCTi₂, GeCV₂, InCHf₂, InCNb₂, InCTi₂, InCZr₂, InNTi₂, InNZr₂, PbCHf₂, PbCTi₂, PbCZr₂, SCNb₂, SCTi₂, SCZr₂, SCZr₂, SeCZr₂, SnCHf₂, SnCNb₂, SnCZr₂, TlCHf₂, TlCTi₂, TlCZr₂

- Delafossite appears in two forms which differ in the stacking of the layers: rhombohedral, prototype CuFeO₂, and hexagonal, shown here. Most of the structures found in the hexagonal phase can also be found in the rhombohedral structure (Marquardt, 2006).
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Hexagonal primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
 \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
 \mathbf{a}_3 &= c \hat{\mathbf{z}}
 \end{aligned}$$


Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	0	$=$	0	(2a)	Al I
$\mathbf{B}_2$	$\frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(2a)	Al I
$\mathbf{B}_3$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(2c)	Cu I
$\mathbf{B}_4$	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(2c)	Cu I
$\mathbf{B}_5$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(4f)	O I
$\mathbf{B}_6$	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	O I
$\mathbf{B}_7$	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(4f)	O I
$\mathbf{B}_8$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	O I

References

- [1] B. U. Köhler and M. Jansen, *Darstellung und Strukturdaten von „Delafossiten“ CuMO₂ (M = Al, Ga, Sc, Y)*, Z. Anorganische und Allgemeine Chemie **543**, 73–80 (1986), doi:10.1002/zaac.19865431209.

Found in

- [1] M. A. Marquardt, N. A. Ashmore, and D. P. Cann, *Crystal chemistry and electrical properties of the delafossite structure*, Thin Solid Films **496**, 146–156 (2006), doi:10.1016/j.tsf.2005.08.316.

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