

CdP₄ Structure:

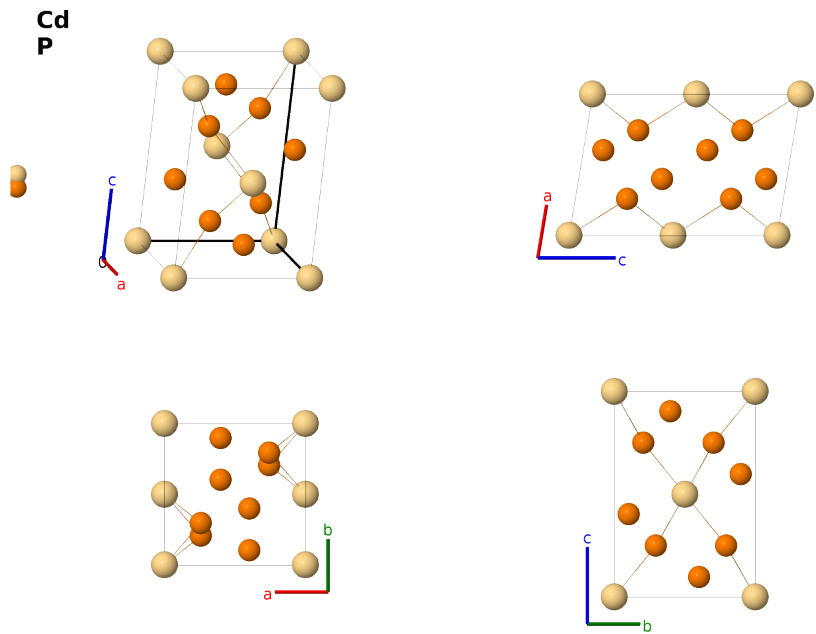
AB4_mP10_14_a_2e-001

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<https://aflow.org/prototype-encyclopedia/L3ZV>

https://aflow.org/prototype-encyclopedia/AB4_mP10_14_a_2e-001



Prototype	CdP ₄
AFLOW prototype label	AB4_mP10_14_a_2e-001
ICSD	25605
CCDC	1601259
Pearson symbol	mP10
Space group number	14
Space group symbol	$P2_1/c$
AFLOW prototype command	<code>aflow --proto=AB4_mP10_14_a_2e-001</code> <code>--params=a, b/a, c/a, β, x_2, y_2, z_2, x_3, y_3, z_3</code>

Other compounds with this structure

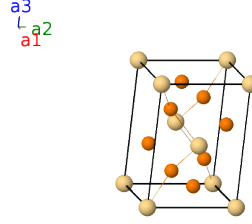
MgP₄, OsP₄, RuP₄

- (Krebs, 1956) give distances in kX. We follow (Wood, 1947) and convert this to Ångstroms by multiplying their distances by 1.00202.

- This is the low temperature form OsP₄ and RuP₄ (Flörke, 1982). For the high temperature form see the CrP₄ structure.

Simple Monoclinic primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \cos \beta \hat{\mathbf{x}} + c \sin \beta \hat{\mathbf{z}} \end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	= 0	=	0	(2a)	Cd I
$\mathbf{B}_2$	= $\frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2} c \cos \beta \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} + \frac{1}{2} c \sin \beta \hat{\mathbf{z}}$	(2a)	Cd I
$\mathbf{B}_3$	= $x_2 \mathbf{a}_1 + y_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	=	$(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} + by_2 \hat{\mathbf{y}} + cz_2 \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_4$	= $-x_2 \mathbf{a}_1 + (y_2 + \frac{1}{2}) \mathbf{a}_2 - (z_2 - \frac{1}{2}) \mathbf{a}_3$	=	$-(ax_2 + c(z_2 - \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} + b(y_2 + \frac{1}{2}) \hat{\mathbf{y}} - c(z_2 - \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_5$	= $-x_2 \mathbf{a}_1 - y_2 \mathbf{a}_2 - z_2 \mathbf{a}_3$	=	$-(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} - by_2 \hat{\mathbf{y}} - cz_2 \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_6$	= $x_2 \mathbf{a}_1 - (y_2 - \frac{1}{2}) \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	=	$(ax_2 + c(z_2 + \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} - b(y_2 - \frac{1}{2}) \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_7$	= $x_3 \mathbf{a}_1 + y_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$	=	$(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} + by_3 \hat{\mathbf{y}} + cz_3 \sin \beta \hat{\mathbf{z}}$	(4e)	P II
$\mathbf{B}_8$	= $-x_3 \mathbf{a}_1 + (y_3 + \frac{1}{2}) \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$	=	$-(ax_3 + c(z_3 - \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} + b(y_3 + \frac{1}{2}) \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P II
$\mathbf{B}_9$	= $-x_3 \mathbf{a}_1 - y_3 \mathbf{a}_2 - z_3 \mathbf{a}_3$	=	$-(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} - by_3 \hat{\mathbf{y}} - cz_3 \sin \beta \hat{\mathbf{z}}$	(4e)	P II
$\mathbf{B}_{10}$	= $x_3 \mathbf{a}_1 - (y_3 - \frac{1}{2}) \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	=	$(ax_3 + c(z_3 + \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} - b(y_3 - \frac{1}{2}) \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P II

References

- [1] H. Krebs, K.-H. Müller, and G. Zürn, *Über kristallisierte Metallpolyphosphide. I. Darstellung und Struktur des CdP₄*, Z. Anorganische und Allgemeine Chemie **285**, 15–28 (1956), doi:10.1016/0022-5088(82)90210-7.
- [2] E. A. Wood, *The Conversion Factor for kX Units to Angström Units*, J. App. Phys. **18**, 929–930 (1947), doi:10.1063/1.1697570.

Found in

- [1] U. Flörke and W. Jeitscho, *Preparation and properties of new modifications of RuP₄ and OsP₄ with CdP₄-type structure*, J. Less-Common Met. **86**, 247–253 (1982), doi:10.1016/0022-5088(82)90210-7.

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