

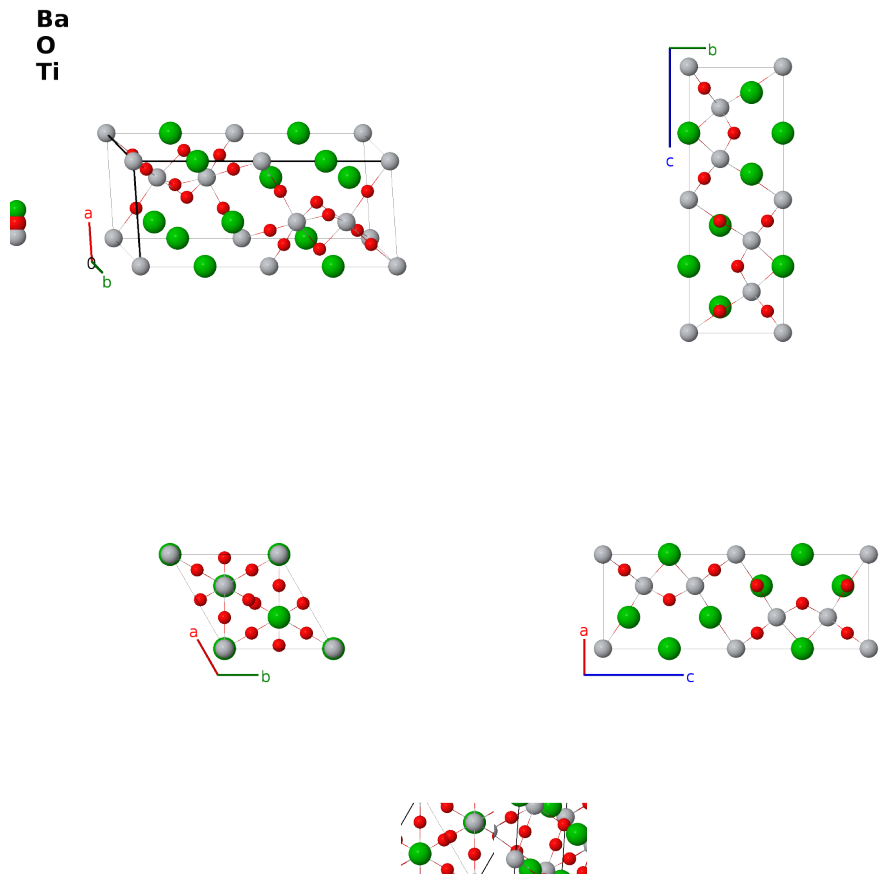
Hexagonal BaTiO₃ Structure: AB3C_hP30_194_bf_hk_af-001

Cite this page as:

- H. Eckert, S. Divilov, M. J. Mehl, D. Hicks, A. C. Zettl, M. Esters, X. Campilongo, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*, Comput. Mater. Sci. **240**, 112988 (2024), doi: 10.1016/j.commatsci.2024.112988.
- D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/prototype-encyclopedia/VHHA>

https://aflow.org/prototype-encyclopedia/AB3C_hP30_194_bf_hk_af-001



Prototype	BaO ₃ Ti
AFLOW prototype label	AB3C_hP30_194_bf_hk_af-001
ICSD	75240
CCDC	1635896
Pearson symbol	hP30
Space group number	194
Space group symbol	$P6_3/mmc$
AFLOW prototype command	<code>aflow --proto=AB3C_hP30_194_bf_hk_af-001</code> <code>--params=a, c/a, z₃, z₄, x₅, x₆, z₆</code>

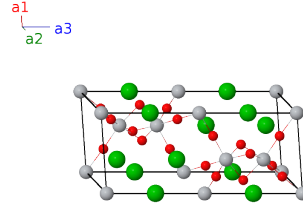
Other compounds with this structure

BaFeO₃, BaMnO₃, BaRuO₃, Ba(Co_xTi_{1-x})O₃, Ba(Cr_xTi_{1-x})O₃, Ba(Fe_xTi_{1-x})O₃, Ba(Ir_xTi_{1-x})O₃, Ba(Mn_xTi_{1-x})O₃, Ba(Os_xTi_{1-x})O₃, Ba(Pd_xTi_{1-x})O₃, Ba(Pt_xTi_{1-x})O₃, Ba(Rh_xTi_{1-x})O₃, Ba(Ru_xTi_{1-x})O₃, Ba(V_xTi_{1-x})O₃

- The perovskite BaTiO₃ undergoes a variety of temperature driven phase transitions. (Shirane, 1957) The first three structures are ferroelectric:
 - Below 193K the structure is rhombohedral.
 - Between 193K and 278K the structure is orthorhombic.
 - Between 278K and 393K the structure is tetragonal. This is the room-temperature form of the material.
 - Above 393K the compound is a cubic perovskite ($E2_1$).
 - Hexagonal BaTiO₃ (this structure) can be stabilized by alloying the titanium sites with other transition metals. (Dickson, 1961) The pure structure has been grown at 1853K and cooled to room temperature. (Akimo, 1994) In some cases there is a deficiency of oxygen atoms. (Dickson, 1961)
 - This pure sample was grown at 1853K and cooled to room temperature.
 - If the transition metal atoms have the composition M1M2₂ and are ordered on the (2a) and (4f) sites, this becomes the hexagonal Ba₃CoIr₂O₉ structure.
 - The ICSD lists BeFeO_{3-x} as the prototype.
-

Hexagonal primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}} \end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= 0$	$=$	0	(2a)	Ti I
$\mathbf{B}_2$	$= \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(2a)	Ti I
$\mathbf{B}_3$	$= \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{4}c \hat{\mathbf{z}}$	(2b)	Ba I
$\mathbf{B}_4$	$= \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{3}{4}c \hat{\mathbf{z}}$	(2b)	Ba I
$\mathbf{B}_5$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(4f)	Ba II
$\mathbf{B}_6$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Ba II
$\mathbf{B}_7$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(4f)	Ba II
$\mathbf{B}_8$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Ba II
$\mathbf{B}_9$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(4f)	Ti II
$\mathbf{B}_{10}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_4 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Ti II
$\mathbf{B}_{11}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - cz_4 \hat{\mathbf{z}}$	(4f)	Ti II

$\mathbf{B}_{12} =$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 - (z_4 - \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6} a \hat{\mathbf{y}} - c (z_4 - \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Ti II
$\mathbf{B}_{13} =$	$x_5 \mathbf{a}_1 + 2x_5 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{3}{2} ax_5 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2} ax_5 \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(6h)	O I
$\mathbf{B}_{14} =$	$-2x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2} ax_5 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2} ax_5 \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(6h)	O I
$\mathbf{B}_{15} =$	$x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\sqrt{3} ax_5 \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(6h)	O I
$\mathbf{B}_{16} =$	$-x_5 \mathbf{a}_1 - 2x_5 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2} ax_5 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2} ax_5 \hat{\mathbf{y}} + \frac{3}{4} c \hat{\mathbf{z}}$	(6h)	O I
$\mathbf{B}_{17} =$	$2x_5 \mathbf{a}_1 + x_5 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{3}{2} ax_5 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2} ax_5 \hat{\mathbf{y}} + \frac{3}{4} c \hat{\mathbf{z}}$	(6h)	O I
$\mathbf{B}_{18} =$	$-x_5 \mathbf{a}_1 + x_5 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\sqrt{3} ax_5 \hat{\mathbf{y}} + \frac{3}{4} c \hat{\mathbf{z}}$	(6h)	O I
$\mathbf{B}_{19} =$	$x_6 \mathbf{a}_1 + 2x_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$\frac{3}{2} ax_6 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{20} =$	$-2x_6 \mathbf{a}_1 - x_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$-\frac{3}{2} ax_6 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{21} =$	$x_6 \mathbf{a}_1 - x_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$-\sqrt{3} ax_6 \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{22} =$	$-x_6 \mathbf{a}_1 - 2x_6 \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	$=$	$-\frac{3}{2} ax_6 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} + c (z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{23} =$	$2x_6 \mathbf{a}_1 + x_6 \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{3}{2} ax_6 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} + c (z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{24} =$	$-x_6 \mathbf{a}_1 + x_6 \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\sqrt{3} ax_6 \hat{\mathbf{y}} + c (z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{25} =$	$2x_6 \mathbf{a}_1 + x_6 \mathbf{a}_2 - z_6 \mathbf{a}_3$	$=$	$\frac{3}{2} ax_6 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} - cz_6 \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{26} =$	$-x_6 \mathbf{a}_1 - 2x_6 \mathbf{a}_2 - z_6 \mathbf{a}_3$	$=$	$-\frac{3}{2} ax_6 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} - cz_6 \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{27} =$	$-x_6 \mathbf{a}_1 + x_6 \mathbf{a}_2 - z_6 \mathbf{a}_3$	$=$	$\sqrt{3} ax_6 \hat{\mathbf{y}} - cz_6 \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{28} =$	$-2x_6 \mathbf{a}_1 - x_6 \mathbf{a}_2 - (z_6 - \frac{1}{2}) \mathbf{a}_3$	$=$	$-\frac{3}{2} ax_6 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} - c (z_6 - \frac{1}{2}) \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{29} =$	$x_6 \mathbf{a}_1 + 2x_6 \mathbf{a}_2 - (z_6 - \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{3}{2} ax_6 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2} ax_6 \hat{\mathbf{y}} - c (z_6 - \frac{1}{2}) \hat{\mathbf{z}}$	(12k)	O II
$\mathbf{B}_{30} =$	$x_6 \mathbf{a}_1 - x_6 \mathbf{a}_2 - (z_6 - \frac{1}{2}) \mathbf{a}_3$	$=$	$-\sqrt{3} ax_6 \hat{\mathbf{y}} - c (z_6 - \frac{1}{2}) \hat{\mathbf{z}}$	(12k)	O II

References

- [1] G. Shirane, H. Danner, and R. Pepinsky, *Neutron Diffraction Study of Orthorhombic BaTiO₃*, Phys. Rev. **105**, 856–860 (1957), doi:10.1103/PhysRev.105.856.
- [2] J. G. Dickson, L. Katz, and R. Ward, *Compounds with the Hexagonal Barium Titanate Structure*, J. Am. Chem. Soc. **83**, 3026–3029 (1961), doi:10.1021/ja01475a012.
- [3] A. W. Hewat, *Structure of rhombohedral ferroelectric barium titanate*, Ferroelectrics **6**, 215–218 (1974), doi:10.1080/00150197408243970.
- [4] J. Akimoto, Y. Gotoh, and Y. Oosawa, *Refinement of Hexagonal BaTiO₃*, Acta Crystallogr. Sect. C **50**, 160–161 (1994), doi:10.1107/S0108270193008637.

First cited in

H. Eckert, S. Divilov, M. J. Mehl, D. Hicks, A. C. Zettel, M. Esters, X. Campilongo, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*, Comput. Mater. Sci. **240**, 112988 (2024), doi: 10.1016/j.commatsci.2024.112988.