

MgZn₂ Hexagonal Laves (C14) Structure: AB2_hP12_194_f_ah-001

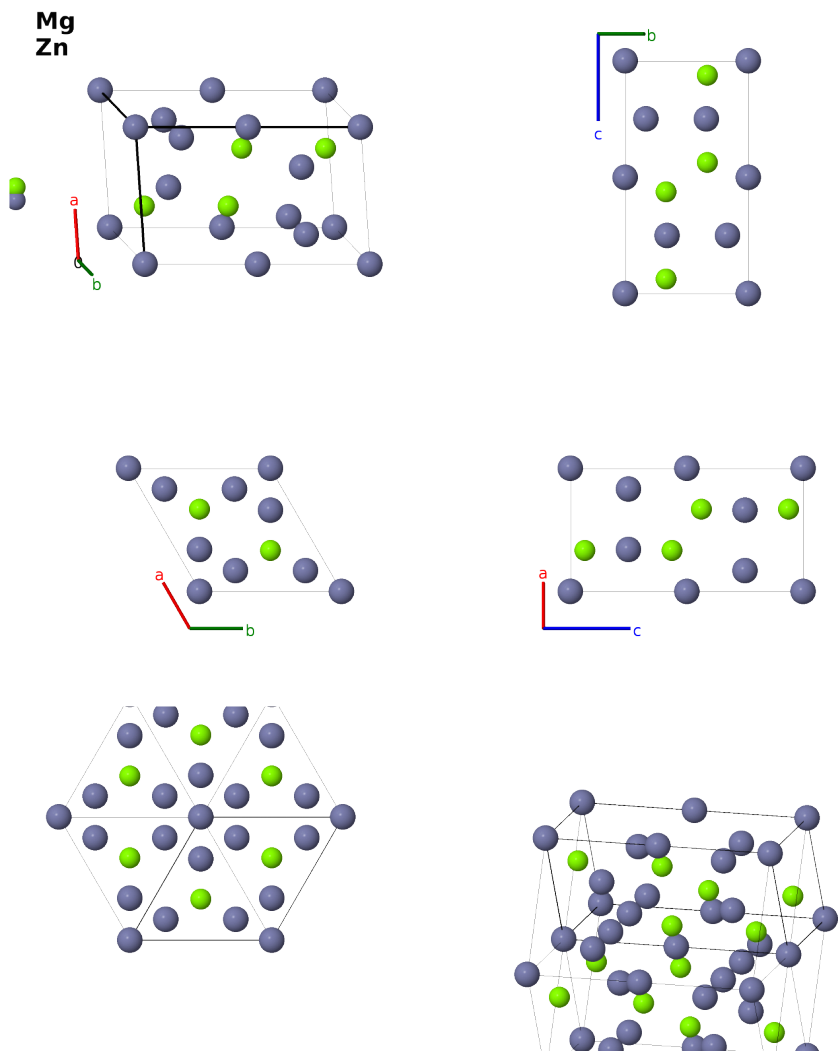
This structure previously used the label(s) **AB2_hP12_194_f_ah**. Calls to these addresses will be redirected here.

Cite this page as:

- H. Eckert, S. Divilov, M. J. Mehl, D. Hicks, A. C. Zettel, M. Esters, X. Campilongo, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*, Comput. Mater. Sci. **240**, 112988 (2024), doi: 10.1016/j.commatsci.2024.112988.
- D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/prototype-encyclopedia/LL0C>

https://aflow.org/prototype-encyclopedia/AB2_hP12_194_f_ah-001



Prototype	MgZn ₂
AFLOW prototype label	AB2_hP12_194_f_ah-001
Strukturbericht designation	C14
Mineral name	Hexagonal Laves

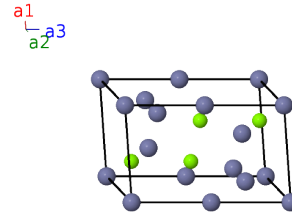
ICSD	46006
CCDC	1614392
Pearson symbol	hP12
Space group number	194
Space group symbol	$P6_3/mmc$
AFLOW prototype command	<code>aflow --proto=AB2_hP12_194_f_ah-001</code> <code>--params=a,c/a,z₂,x₃</code>

Other compounds with this structure

BaMg₂, CaCd₂, CaLi₂, CaMg₂, CdCu₂, CrBe₂, DyOs₂, DyRu₂, DyTc₂, ErMg₂, ErMn₂, ErOs₂, ErRu₂, ErTc₂, FeBe₂, GdOs₂, GdRu₂, GdTc₂, HfCr₂, HfFe₂ (HT), HfMn₂ (HT), HfRe₂, HoMg₂, HoOs₂, HoRe₂, HoRu₂, HoTc₂, KNa₂, KPb₂, LuMn₂, LuOs₂, LuRe₂, LuRh₂, LuRu₂, LuTe₂, MgZn₂, MnBe₂, MoBe₂, MoFe₂, NbCr₂ (HT), NbFe₂, NbMn₂, NdMn₂, NdOs₂, PrMn₂, PrOs₂, PuOs₂, ReBe₂, RuBe₂, ScMn₂, ScOs₂, ScRe₂, ScRu₂, ScTc₂, SmOs₂, SrMg₂, α -TaCo₂ (LT), TaCr₂ (HT), TaFe₂, TaMn₂, TbOs₂, TbRe₂, TbRu₂, TbTc₂, ThMn₂, ThRe₂, TiCr₂ (HT), TiFe₂, TiMn₂, TiZn₂, TmMn₂, TmRu₂, TmTc₂, UNi₂, URe₂ (HT), URe₂, VBe₂, WBe₂, WFe₂, YOs₂, YRe₂, YRu₂, YTc₂, ZrCr₂, ZrMn₂, ZrOs₂, ZrRe₂, ZrRu₂ (HT), ZrTc₂

- U₂Cr₃Si is a ternary form of this structure.

Hexagonal primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
 \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
 \mathbf{a}_3 &= c \hat{\mathbf{z}}
 \end{aligned}$$


Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= 0$	$=$	0	(2a)	Zn I
$\mathbf{B}_2$	$= \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(2a)	Zn I
$\mathbf{B}_3$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_2 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_2 \hat{\mathbf{z}}$	(4f)	Mg I
$\mathbf{B}_4$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Mg I
$\mathbf{B}_5$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_2 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - cz_2 \hat{\mathbf{z}}$	(4f)	Mg I
$\mathbf{B}_6$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 - (z_2 - \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - c(z_2 - \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Mg I
$\mathbf{B}_7$	$= x_3 \mathbf{a}_1 + 2x_3 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_3 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h)	Zn II
$\mathbf{B}_8$	$= -2x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_3 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h)	Zn II
$\mathbf{B}_9$	$= x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\sqrt{3}ax_3 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h)	Zn II
$\mathbf{B}_{10}$	$= -x_3 \mathbf{a}_1 - 2x_3 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_3 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h)	Zn II
$\mathbf{B}_{11}$	$= 2x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_3 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h)	Zn II
$\mathbf{B}_{12}$	$= -x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\sqrt{3}ax_3 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h)	Zn II

References

- [1] T. Ohba, Y. Kitano, and Y. Komura, *The charge-density study of the Laves phases, $MgZn_2$ and $MgCu_2$* , Acta Crystallogr. Sect. C **40**, 1–5 (1984), doi:10.1107/S0108270184002791.

First cited in

M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. L. W. Hart, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 1*, Comput. Mater. Sci. **136**, S1-828 (2017). doi: 10.1016/j.commatsci.2017.01.017