

$C26_a$ (NO_2) (*Obsolete*) Structure: AB2_cI36_199_b_c-001

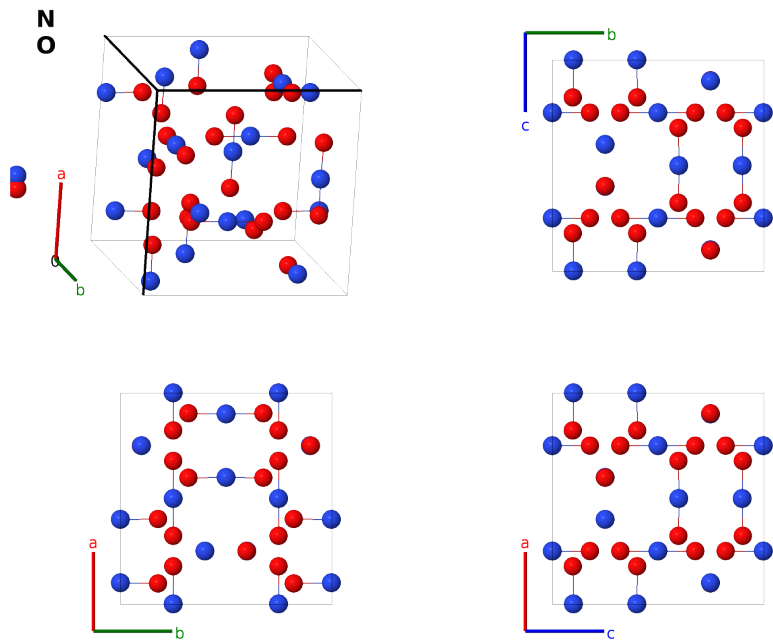
This structure previously used the label(s) **AB2_cI36_199_b_c**. Calls to these addresses will be redirected here.

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<https://aflow.org/prototype-encyclopedia/VGMG>

https://aflow.org/prototype-encyclopedia/AB2_cI36_199_b_c-001

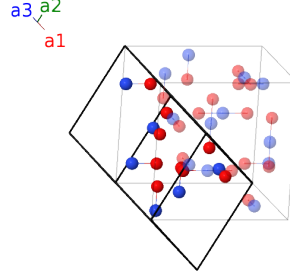


Prototype	NO_2
AFLOW prototype label	AB2_cI36_199_b_c-001
<i>Strukturbericht</i> designation	$C26_a$
ICSD	31175
CCDC	1605327
Pearson symbol	cI36
Space group number	199
Space group symbol	$I2_13$
AFLOW prototype command	<code>aflow --proto=AB2_cI36_199_b_c-001</code> <code>--params=a, x_1, x_2, y_2, z_2</code>

- (Hermann, 1937) listed two possible structures for the low temperature solid cubic phase of NO₂, which were given *Strukturbericht* designations *C26_a* and *C26_b*, the only structures with Roman subscripts in the original series.
- *C26_a* (AB2_cI36_199_b.c) was set in space group *I2₁3* #199. Hermann noted that this structure has a very short distance (1.88Å) between oxygen atoms on different NO₂ molecules, and that this structure does not form the (NO₂)₂ aggregate molecule found in the *C26_b* structure, making “making this proposed structure very unlikely.”
- Recognizing this, (Hendricks, 1931) suggested that NO₂ was actually in space group *I23* #197. (Hermann, 1997) gave this structure the *C26_b* designation, but noted that based on Hendricks’s atomic positions the space group was actually *Im* $\bar{3}$ #204.
- The modern accepted structure for NO₂, AB2_cI36_204_d.g, is set in space group *Im* $\bar{3}$, confirming Hendricks.

Body-centered Cubic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}a\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{1}{4}\mathbf{a}_1 + (x_1 + \frac{1}{4})\mathbf{a}_2 + x_1\mathbf{a}_3 =$	$ax_1\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{z}}$	(12b)	N I
$\mathbf{B}_2$	$= \frac{3}{4}\mathbf{a}_1 - (x_1 - \frac{1}{4})\mathbf{a}_2 - (x_1 - \frac{1}{2})\mathbf{a}_3 =$	$-ax_1\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{4}a\hat{\mathbf{z}}$	(12b)	N I
$\mathbf{B}_3$	$= x_1\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + (x_1 + \frac{1}{4})\mathbf{a}_3 =$	$\frac{1}{4}a\hat{\mathbf{x}} + ax_1\hat{\mathbf{y}}$	(12b)	N I
$\mathbf{B}_4$	$= -(x_1 - \frac{1}{2})\mathbf{a}_1 + \frac{3}{4}\mathbf{a}_2 - (x_1 - \frac{1}{4})\mathbf{a}_3 =$	$\frac{1}{4}a\hat{\mathbf{x}} - ax_1\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}}$	(12b)	N I
$\mathbf{B}_5$	$= (x_1 + \frac{1}{4})\mathbf{a}_1 + x_1\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3 =$	$\frac{1}{4}a\hat{\mathbf{y}} + ax_1\hat{\mathbf{z}}$	(12b)	N I
$\mathbf{B}_6$	$= -(x_1 - \frac{1}{4})\mathbf{a}_1 - (x_1 - \frac{1}{2})\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3 =$	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{y}} - ax_1\hat{\mathbf{z}}$	(12b)	N I
$\mathbf{B}_7$	$= (y_2 + z_2)\mathbf{a}_1 + (x_2 + z_2)\mathbf{a}_2 + (x_2 + y_2)\mathbf{a}_3 =$	$ax_2\hat{\mathbf{x}} + ay_2\hat{\mathbf{y}} + az_2\hat{\mathbf{z}}$	(24c)	O I
$\mathbf{B}_8$	$= (-y_2 + z_2 + \frac{1}{2})\mathbf{a}_1 - (x_2 - z_2)\mathbf{a}_2 - (x_2 + y_2 - \frac{1}{2})\mathbf{a}_3 =$	$-ax_2\hat{\mathbf{x}} - a(y_2 - \frac{1}{2})\hat{\mathbf{y}} + az_2\hat{\mathbf{z}}$	(24c)	O I
$\mathbf{B}_9$	$= (y_2 - z_2)\mathbf{a}_1 - (x_2 + z_2 - \frac{1}{2})\mathbf{a}_2 + (-x_2 + y_2 + \frac{1}{2})\mathbf{a}_3 =$	$-a(x_2 - \frac{1}{2})\hat{\mathbf{x}} + ay_2\hat{\mathbf{y}} - az_2\hat{\mathbf{z}}$	(24c)	O I
$\mathbf{B}_{10}$	$= -(y_2 + z_2 - \frac{1}{2})\mathbf{a}_1 + (x_2 - z_2 + \frac{1}{2})\mathbf{a}_2 + (x_2 - y_2)\mathbf{a}_3 =$	$ax_2\hat{\mathbf{x}} - ay_2\hat{\mathbf{y}} - a(z_2 - \frac{1}{2})\hat{\mathbf{z}}$	(24c)	O I
$\mathbf{B}_{11}$	$= (x_2 + y_2)\mathbf{a}_1 + (y_2 + z_2)\mathbf{a}_2 + (x_2 + z_2)\mathbf{a}_3 =$	$az_2\hat{\mathbf{x}} + ax_2\hat{\mathbf{y}} + ay_2\hat{\mathbf{z}}$	(24c)	O I
$\mathbf{B}_{12}$	$= -(x_2 + y_2 - \frac{1}{2})\mathbf{a}_1 + (-y_2 + z_2 + \frac{1}{2})\mathbf{a}_2 - (x_2 - z_2)\mathbf{a}_3 =$	$az_2\hat{\mathbf{x}} - ax_2\hat{\mathbf{y}} - a(y_2 - \frac{1}{2})\hat{\mathbf{z}}$	(24c)	O I
$\mathbf{B}_{13}$	$= (-x_2 + y_2 + \frac{1}{2})\mathbf{a}_1 + (y_2 - z_2)\mathbf{a}_2 - (x_2 + z_2 - \frac{1}{2})\mathbf{a}_3 =$	$-az_2\hat{\mathbf{x}} - a(x_2 - \frac{1}{2})\hat{\mathbf{y}} + ay_2\hat{\mathbf{z}}$	(24c)	O I

$$\mathbf{B}_{14} = (x_2 - y_2) \mathbf{a}_1 - (y_2 + z_2 - \frac{1}{2}) \mathbf{a}_2 + (x_2 - z_2 + \frac{1}{2}) \mathbf{a}_3 = -a (z_2 - \frac{1}{2}) \hat{\mathbf{x}} + ax_2 \hat{\mathbf{y}} - ay_2 \hat{\mathbf{z}} \quad (24c) \quad \text{O I}$$

$$\mathbf{B}_{15} = (x_2 + z_2) \mathbf{a}_1 + (x_2 + y_2) \mathbf{a}_2 + (y_2 + z_2) \mathbf{a}_3 = ay_2 \hat{\mathbf{x}} + az_2 \hat{\mathbf{y}} + ax_2 \hat{\mathbf{z}} \quad (24c) \quad \text{O I}$$

$$\mathbf{B}_{16} = -(x_2 - z_2) \mathbf{a}_1 - (x_2 + y_2 - \frac{1}{2}) \mathbf{a}_2 + (-y_2 + z_2 + \frac{1}{2}) \mathbf{a}_3 = -a (y_2 - \frac{1}{2}) \hat{\mathbf{x}} + az_2 \hat{\mathbf{y}} - ax_2 \hat{\mathbf{z}} \quad (24c) \quad \text{O I}$$

$$\mathbf{B}_{17} = -(x_2 + z_2 - \frac{1}{2}) \mathbf{a}_1 + (-x_2 + y_2 + \frac{1}{2}) \mathbf{a}_2 + (y_2 - z_2) \mathbf{a}_3 = ay_2 \hat{\mathbf{x}} - az_2 \hat{\mathbf{y}} - a (x_2 - \frac{1}{2}) \hat{\mathbf{z}} \quad (24c) \quad \text{O I}$$

$$\mathbf{B}_{18} = (x_2 - z_2 + \frac{1}{2}) \mathbf{a}_1 + (x_2 - y_2) \mathbf{a}_2 - (y_2 + z_2 - \frac{1}{2}) \mathbf{a}_3 = -ay_2 \hat{\mathbf{x}} - a (z_2 - \frac{1}{2}) \hat{\mathbf{y}} + ax_2 \hat{\mathbf{z}} \quad (24c) \quad \text{O I}$$

References

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- [2] P. Villars and K. Cenzual, eds., *Crystal Structure Data of Inorganic Compounds*, vol. III (Springer-Verlag, Berlin, Heidelberg, 2005).
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First cited in

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