

Na₂HgO₂ Structure:

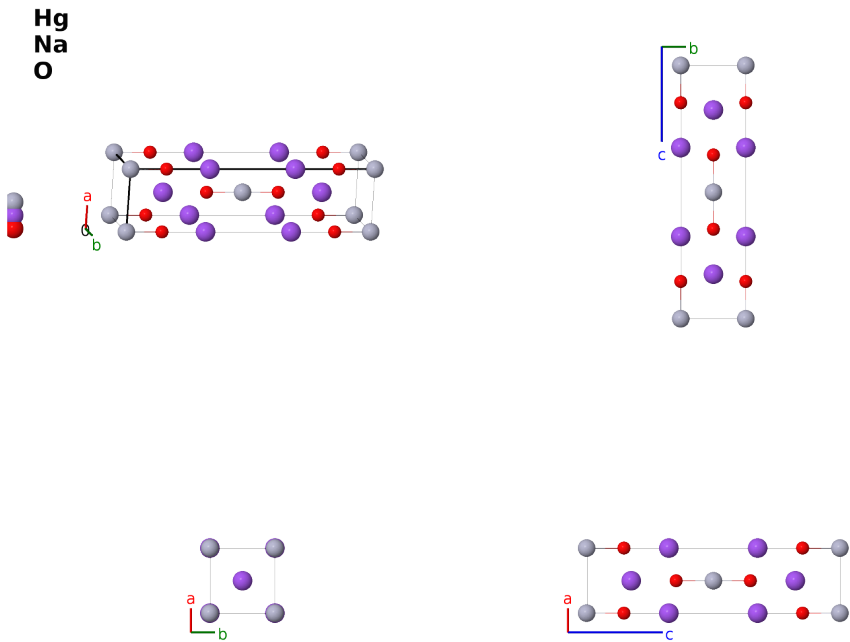
AB2C2_tI10_139_a_e_e-001

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- D. Hicks, M. J. Mehl, M. Esters, C. Osos, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/prototype-encyclopedia/6LGV>

https://aflow.org/prototype-encyclopedia/AB2C2_tI10_139_a_e_e-001



Prototype	HgNa ₂ O ₂
AFLOW prototype label	AB2C2_tI10_139_a_e_e-001
ICSD	25511
CCDC	1601167
Pearson symbol	tI10
Space group number	139
Space group symbol	<i>I4/mmm</i>
AFLOW prototype command	<code>aflow --proto=AB2C2_tI10_139_a_e_e-001 --params=a, c/a, z₂, z₃</code>

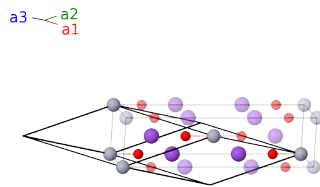
Other compounds with this structure

Cs₂HgO₂, Li₂HgO₂, Rb₂HgO₂, U₂IrC₂, U₂RuC₂

- (Hoppe, 1964) give the space group as $I422$ #97, but the Wyckoff positions they use are consistent with $I4/mmm$ #139, so we use the higher symmetry space group.

Body-centered Tetragonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	0	$=$	0	(2a)	Hg I
$\mathbf{B}_2$	$z_2 \mathbf{a}_1 + z_2 \mathbf{a}_2$	$=$	$cz_2 \hat{\mathbf{z}}$	(4e)	Na I
$\mathbf{B}_3$	$-z_2 \mathbf{a}_1 - z_2 \mathbf{a}_2$	$=$	$-cz_2 \hat{\mathbf{z}}$	(4e)	Na I
$\mathbf{B}_4$	$z_3 \mathbf{a}_1 + z_3 \mathbf{a}_2$	$=$	$cz_3 \hat{\mathbf{z}}$	(4e)	O I
$\mathbf{B}_5$	$-z_3 \mathbf{a}_1 - z_3 \mathbf{a}_2$	$=$	$-cz_3 \hat{\mathbf{z}}$	(4e)	O I

References

- [1] R. Hoppe and H.-J. Rohrborn, *Oxomercurate(II) der Alkalimetalle*, M_2HgO_2 , Zeitschrift für anorganische und allgemeine Chemie **329**, 110–122 (1964), doi:10.1002/zaac.19643290115.

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