

# NaSbF<sub>4</sub>(OH)<sub>2</sub> (*J*1<sub>12</sub>) Structure: A6BC\_hP16\_163\_i\_b\_c-001

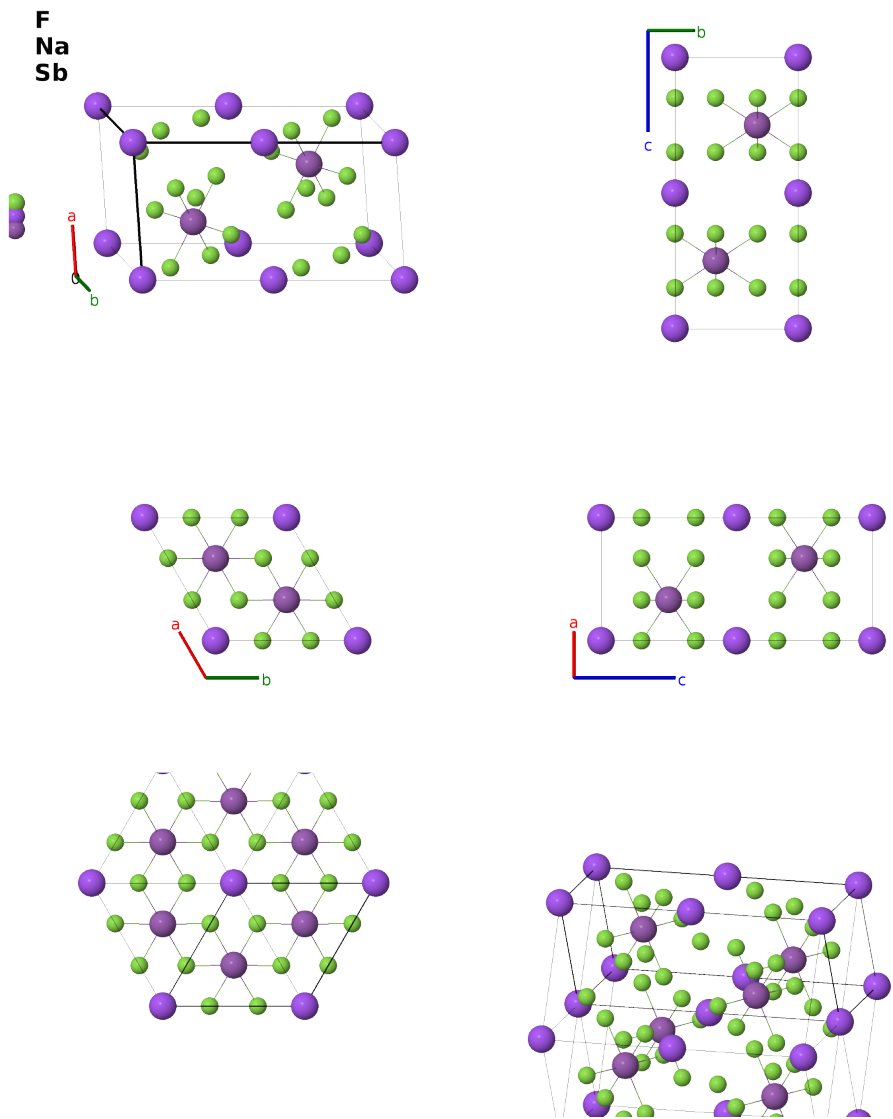
This structure previously used the label(s) **A6BC\_hP16\_163\_i\_b\_c**. Calls to these addresses will be redirected here.

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<https://aflow.org/prototype-encyclopedia/KLNT>

[https://aflow.org/prototype-encyclopedia/A6BC\\_hP16\\_163\\_i\\_b\\_c-001](https://aflow.org/prototype-encyclopedia/A6BC_hP16_163_i_b_c-001)



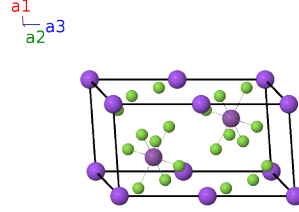
|                             |                          |
|-----------------------------|--------------------------|
| Prototype                   | F <sub>6</sub> NaSb      |
| AFLOW prototype label       | A6BC_hP16_163_i_b_c-001  |
| Strukturbericht designation | <i>J</i> 1 <sub>12</sub> |

|                         |                                                                                                                                 |
|-------------------------|---------------------------------------------------------------------------------------------------------------------------------|
| ICSD                    | 25539                                                                                                                           |
| CCDC                    | 1601193                                                                                                                         |
| Pearson symbol          | hP16                                                                                                                            |
| Space group number      | 163                                                                                                                             |
| Space group symbol      | $P\bar{3}1c$                                                                                                                    |
| AFLOW prototype command | <code>aflow --proto=A6BC_hP16_163_i_b_c-001</code><br><code>--params=a, c/a, x<sub>3</sub>, y<sub>3</sub>, z<sub>3</sub></code> |

- The (2i) site, which we show as occupied by fluorine, is actually occupied by a random mixture of fluorine atoms (67%) and OH radicals (33%).
- Although the replacement of fluorine by OH does not affect the shape of the Sb-(F,OH)<sub>6</sub> ions, it has a profound effect on the structure, as can be seen by looking at NaSbF<sub>6</sub> and NaSb(OH)<sub>6</sub> (*J111*).

### Trigonal (Hexagonal) primitive vectors

$$\begin{aligned}
\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
\mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
\mathbf{a}_3 &= c \hat{\mathbf{z}}
\end{aligned}$$



### Basis vectors

|                   | Lattice coordinates                                                                 |     | Cartesian coordinates                                                                                                                 | Wyckoff position | Atom type |
|-------------------|-------------------------------------------------------------------------------------|-----|---------------------------------------------------------------------------------------------------------------------------------------|------------------|-----------|
| $\mathbf{B}_1$    | $= 0$                                                                               | $=$ | $0$                                                                                                                                   | (2b)             | Na I      |
| $\mathbf{B}_2$    | $= \frac{1}{2} \mathbf{a}_3$                                                        | $=$ | $\frac{1}{2}c \hat{\mathbf{z}}$                                                                                                       | (2b)             | Na I      |
| $\mathbf{B}_3$    | $= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$  | $=$ | $\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$                                | (2c)             | Sb I      |
| $\mathbf{B}_4$    | $= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$  | $=$ | $\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$                                | (2c)             | Sb I      |
| $\mathbf{B}_5$    | $= x_3 \mathbf{a}_1 + y_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$                          | $=$ | $\frac{1}{2}a(x_3 + y_3) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a(x_3 - y_3) \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$                  | (12i)            | F I       |
| $\mathbf{B}_6$    | $= -y_3 \mathbf{a}_1 + (x_3 - y_3) \mathbf{a}_2 + z_3 \mathbf{a}_3$                 | $=$ | $\frac{1}{2}a(x_3 - 2y_3) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$                         | (12i)            | F I       |
| $\mathbf{B}_7$    | $= -(x_3 - y_3) \mathbf{a}_1 - x_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$                 | $=$ | $-\frac{1}{2}a(2x_3 - y_3) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ay_3 \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$                        | (12i)            | F I       |
| $\mathbf{B}_8$    | $= -y_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$         | $=$ | $-\frac{1}{2}a(x_3 + y_3) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a(x_3 - y_3) \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \hat{\mathbf{z}}$ | (12i)            | F I       |
| $\mathbf{B}_9$    | $= -(x_3 - y_3) \mathbf{a}_1 + y_3 \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$ | $=$ | $\frac{1}{2}a(-x_3 + 2y_3) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \hat{\mathbf{z}}$        | (12i)            | F I       |
| $\mathbf{B}_{10}$ | $= x_3 \mathbf{a}_1 + (x_3 - y_3) \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$  | $=$ | $\frac{1}{2}a(2x_3 - y_3) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ay_3 \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \hat{\mathbf{z}}$         | (12i)            | F I       |
| $\mathbf{B}_{11}$ | $= -x_3 \mathbf{a}_1 - y_3 \mathbf{a}_2 - z_3 \mathbf{a}_3$                         | $=$ | $-\frac{1}{2}a(x_3 + y_3) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a(x_3 - y_3) \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$                 | (12i)            | F I       |
| $\mathbf{B}_{12}$ | $= y_3 \mathbf{a}_1 - (x_3 - y_3) \mathbf{a}_2 - z_3 \mathbf{a}_3$                  | $=$ | $\frac{1}{2}a(-x_3 + 2y_3) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$                        | (12i)            | F I       |
| $\mathbf{B}_{13}$ | $= (x_3 - y_3) \mathbf{a}_1 + x_3 \mathbf{a}_2 - z_3 \mathbf{a}_3$                  | $=$ | $\frac{1}{2}a(2x_3 - y_3) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ay_3 \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$                         | (12i)            | F I       |

$$\mathbf{B}_{14} = y_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + \left(z_3 + \frac{1}{2}\right) \mathbf{a}_3 = \frac{1}{2}a(x_3 + y_3) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a(x_3 - y_3) \hat{\mathbf{y}} + c\left(z_3 + \frac{1}{2}\right) \hat{\mathbf{z}} \quad (12i) \quad \text{F I}$$

$$\mathbf{B}_{15} = (x_3 - y_3) \mathbf{a}_1 - y_3 \mathbf{a}_2 + \left(z_3 + \frac{1}{2}\right) \mathbf{a}_3 = \frac{1}{2}a(x_3 - 2y_3) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}} + c\left(z_3 + \frac{1}{2}\right) \hat{\mathbf{z}} \quad (12i) \quad \text{F I}$$

$$\mathbf{B}_{16} = -x_3 \mathbf{a}_1 - (x_3 - y_3) \mathbf{a}_2 + \left(z_3 + \frac{1}{2}\right) \mathbf{a}_3 = -\frac{1}{2}a(2x_3 - y_3) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ay_3 \hat{\mathbf{y}} + c\left(z_3 + \frac{1}{2}\right) \hat{\mathbf{z}} \quad (12i) \quad \text{F I}$$

## References

- [1] N. Schrewelius, *Röntgenuntersuchung der Verbindungen NaSb(OH)<sub>6</sub>, NaSbF<sub>6</sub>, NaSbO<sub>3</sub> und gleichartiger Stoffe*, Z. Anorganische und Allgemeine Chemie **238**, 241–254 (1938), doi:10.1002/zaac.19382380209.

## Found in

- [1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).

## First cited in

D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.