

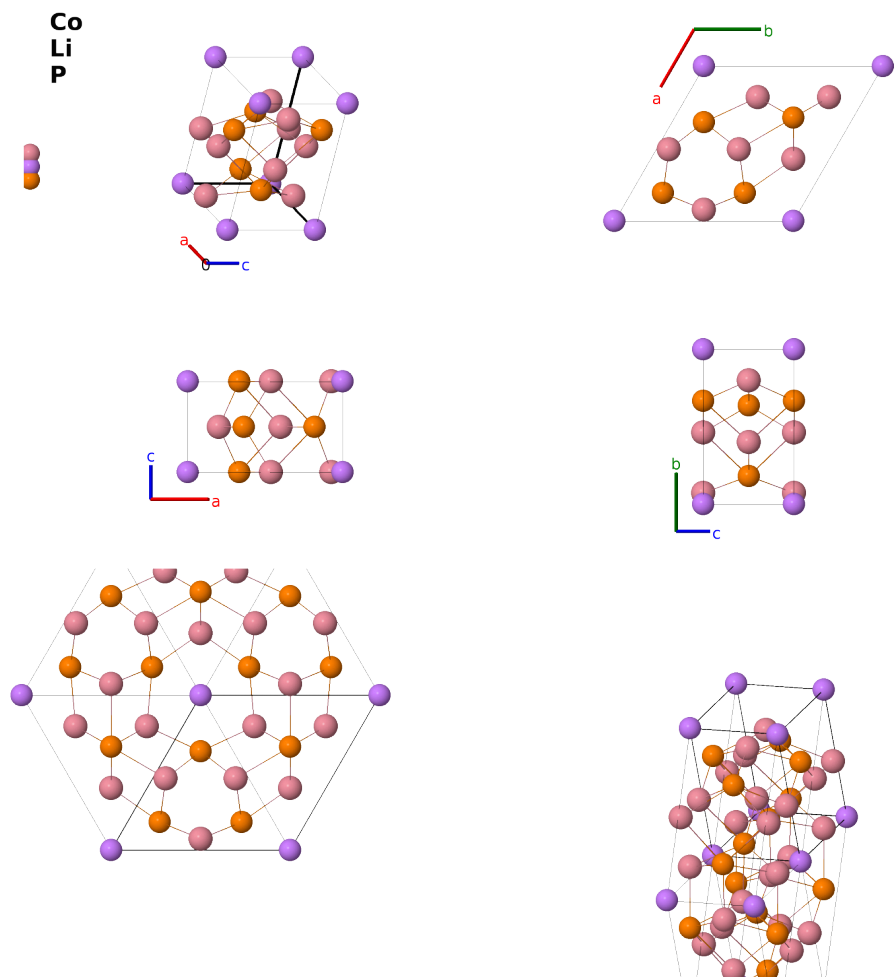
LiCo₆P₄ Structure: A6BC4_hP11_187_jk_a_ck-001

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<https://aflow.org/prototype-encyclopedia/NFL9>

https://aflow.org/prototype-encyclopedia/A6BC4_hP11_187_jk_a_ck-001



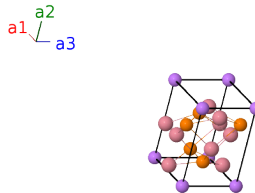
Prototype	Co ₆ LiP ₄
AFLOW prototype label	A6BC4_hP11_187_jk_a_ck-001
ICSD	69692
CCDC	1631543
Pearson symbol	hP11
Space group number	187
Space group symbol	$P\bar{6}m2$

AFLOW prototype command `aflow --proto=A6BC4_hP11_187_jk_a_ck-001`
 `--params=a, c/a, x3, x4, x5`

Other compounds with this structure

CeCo₆P₄, CeRh₆Ge₄, CeRh₆Ge₄, DyCo₆P₄, DyRh₆Ge₄, ErCo₆P₄, ErRh₆Ge₄, EuCo₆P₄, EuRh₆Ge₄, GdCo₆P₄, GdRh₆Ge₄, HoCo₆P₄, HoRh₆Ge₄, LaCo₆P₄, LaRh₆Ge₄, LuCo₆P₄, LuRh₆Ge₄, LuRh₆P₄, NdCo₆P₄, NdRh₆Ge₄, PrCo₆P₄, PrRh₆Ge₄, ScRh₆P₄, SmCo₆P₄, SmRh₆Ge₄, TbCo₆P₄, TbRh₆Ge₄, TmCo₆P₄, TmRh₆Ge₄, YCo₆P₄, YRh₆Ge₄, YRh₆P₄, YbCo₆P₄, YbRh₆Ge₄

Hexagonal primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}} \end{aligned}$$


Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$=$	0	$=$	0	(1a) Li I
$\mathbf{B}_2$	$=$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}}$	(1c) P I
$\mathbf{B}_3$	$=$	$x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2$	$=$	$-\sqrt{3}ax_3 \hat{\mathbf{y}}$	(3j) Co I
$\mathbf{B}_4$	$=$	$x_3 \mathbf{a}_1 + 2x_3 \mathbf{a}_2$	$=$	$\frac{3}{2}ax_3 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}}$	(3j) Co I
$\mathbf{B}_5$	$=$	$-2x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2$	$=$	$-\frac{3}{2}ax_3 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3 \hat{\mathbf{y}}$	(3j) Co I
$\mathbf{B}_6$	$=$	$x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$-\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3k) Co II
$\mathbf{B}_7$	$=$	$x_4 \mathbf{a}_1 + 2x_4 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3k) Co II
$\mathbf{B}_8$	$=$	$-2x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3k) Co II
$\mathbf{B}_9$	$=$	$x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$-\sqrt{3}ax_5 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3k) P II
$\mathbf{B}_{10}$	$=$	$x_5 \mathbf{a}_1 + 2x_5 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_5 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3k) P II
$\mathbf{B}_{11}$	$=$	$-2x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_5 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(3k) P II

References

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Found in

- [1] S. F. Matar, A. Al-Alam, N. Ouaini, and R. Pöttgen, *Ab initio investigations of the electronic structures and chemical bonding in LiCo₆P₄ and Li₂Co₁₂P₇*, J. Solid State Chem. **202**, 227–233 (2013), doi:10.1016/j.jssc.2013.03.032.

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