

Ag₅Pb₂O₆ Structure:

A5B6C2_hP13_157_2ac_2c_b-001

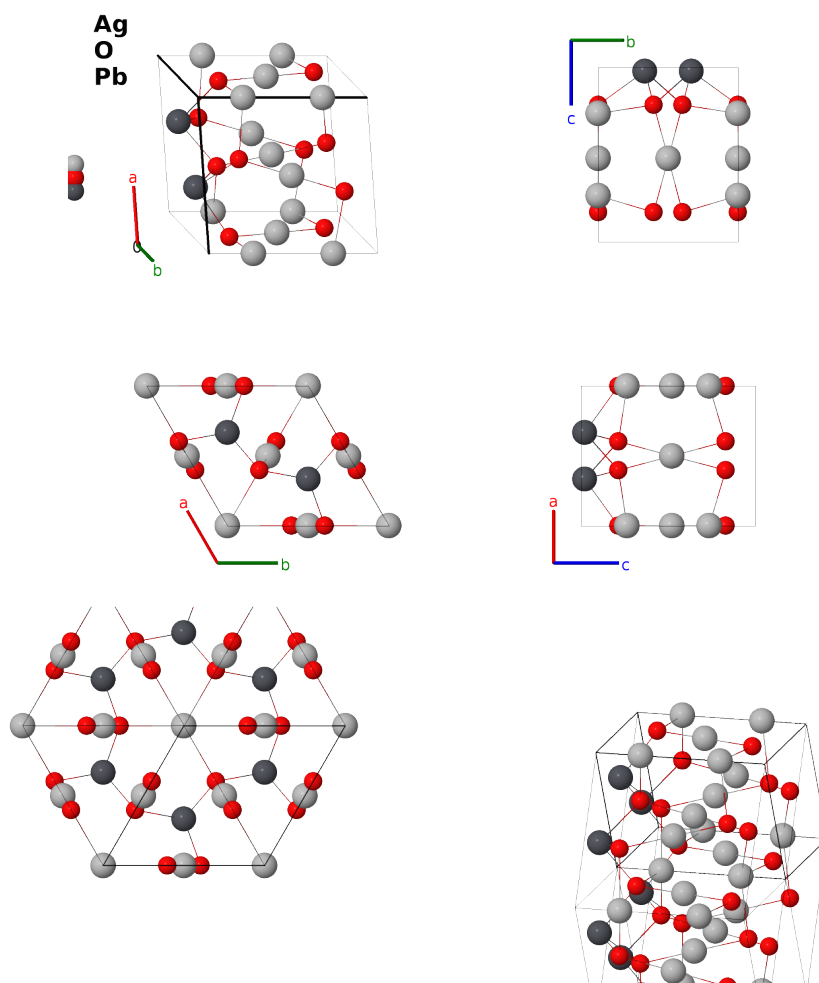
This structure previously used the label(s) **A5B6C2_hP13_157_2ac_2c_b**. Calls to these addresses will be redirected here.

Cite this page as:

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<https://aflow.org/prototype-encyclopedia/1YM0>

https://aflow.org/prototype-encyclopedia/A5B6C2_hP13_157_2ac_2c_b-001



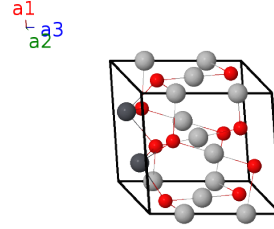
Prototype	Ag ₅ O ₆ Pb ₂
AFLOW prototype label	A5B6C2_hP13_157_2ac_2c_b-001
ICSD	24038
CCDC	1600094
Pearson symbol	hP13

Space group number	157
Space group symbol	$P31m$
AFLOW prototype command	<code>aflow --proto=A5B6C2_hP13_157_2ac_2c_b-001</code> <code>--params=a, c/a, z₁, z₂, z₃, x₄, z₄, x₅, z₅, x₆, z₆</code>

- The original reference (Byström, 1950) lists this structure as $\text{Ag}_5\text{Pb}_2\text{O}_6$, while (Villars, 1985) lists it as Ag_2PbO_3 . (Byström, 1950) provides three Wyckoff positions for Ag: (1a), (1a), and (3c), while (Villars, 1985) only provides two: (1a) and (3c), giving rise to the stoichiometry discrepancy. While both descriptions yield space group $P31m$ #157, we use the structure and coordinates provided by the original reference (Byström, 1950).

Trigonal (Hexagonal) primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_3 &= c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= z_1 \mathbf{a}_3$	$=$	$cz_1 \hat{\mathbf{z}}$	(1a)	Ag I
$\mathbf{B}_2$	$= z_2 \mathbf{a}_3$	$=$	$cz_2 \hat{\mathbf{z}}$	(1a)	Ag II
$\mathbf{B}_3$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + cz_3\hat{\mathbf{z}}$	(2b)	Pb I
$\mathbf{B}_4$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + cz_3\hat{\mathbf{z}}$	(2b)	Pb I
$\mathbf{B}_5$	$= x_4 \mathbf{a}_1 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}ax_4\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + cz_4\hat{\mathbf{z}}$	(3c)	Ag III
$\mathbf{B}_6$	$= x_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}ax_4\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + cz_4\hat{\mathbf{z}}$	(3c)	Ag III
$\mathbf{B}_7$	$= -x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$-ax_4\hat{\mathbf{x}} + cz_4\hat{\mathbf{z}}$	(3c)	Ag III
$\mathbf{B}_8$	$= x_5 \mathbf{a}_1 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}ax_5\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_5\hat{\mathbf{y}} + cz_5\hat{\mathbf{z}}$	(3c)	O I
$\mathbf{B}_9$	$= x_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}ax_5\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5\hat{\mathbf{y}} + cz_5\hat{\mathbf{z}}$	(3c)	O I
$\mathbf{B}_{10}$	$= -x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$-ax_5\hat{\mathbf{x}} + cz_5\hat{\mathbf{z}}$	(3c)	O I
$\mathbf{B}_{11}$	$= x_6 \mathbf{a}_1 + z_6 \mathbf{a}_3$	$=$	$\frac{1}{2}ax_6\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_6\hat{\mathbf{y}} + cz_6\hat{\mathbf{z}}$	(3c)	O II
$\mathbf{B}_{12}$	$= x_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$\frac{1}{2}ax_6\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_6\hat{\mathbf{y}} + cz_6\hat{\mathbf{z}}$	(3c)	O II
$\mathbf{B}_{13}$	$= -x_6 \mathbf{a}_1 - x_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$-ax_6\hat{\mathbf{x}} + cz_6\hat{\mathbf{z}}$	(3c)	O II

References

- [1] A. Byström and L. Evers, *The Crystal Structures of Ag_2PbO_3 and $\text{Ag}_5\text{Pb}_2\text{O}_6$* , Acta Chem. Scand. **4**, 613–627 (1950), doi:10.3891/acta.chem.scand.04-0613.

Found in

- [1] P. Villars and L. D. Calvert, eds., *Pearson's Handbook of Crystallographic Data for Intermetallic Phases*, vol. 1 (American Society of Metals, Materials Park, Ohio, 1985).

First cited in

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