

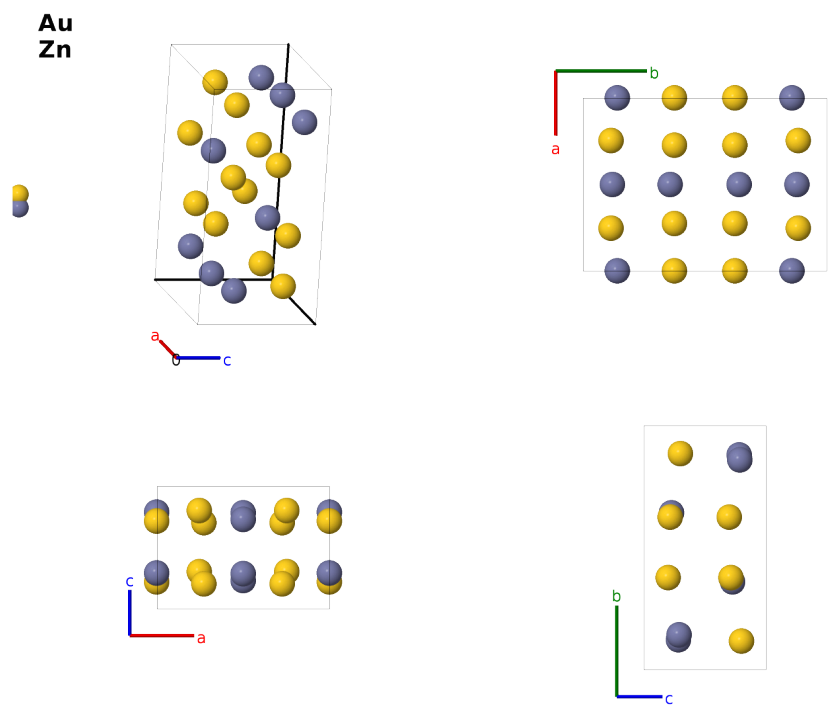
Metastable Au_5Zn_3 Structure: A5B3_oP16_26_a2c_a2b-001

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<https://aflow.org/prototype-encyclopedia/PD7K>

https://aflow.org/prototype-encyclopedia/A5B3_oP16_26_a2c_a2b-001



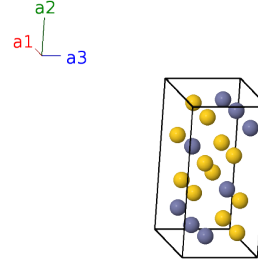
Prototype	Au_5Zn_3
AFLOW prototype label	A5B3_oP16_26_a2c_a2b-001
ICSD	58629
CCDC	1622635
Pearson symbol	oP16
Space group number	26
Space group symbol	$Pmc2_1$
AFLOW prototype command	<code>aflow --proto=A5B3_oP16_26_a2c_a2b-001</code> <code>--params=$a, b/a, c/a, y_1, z_1, y_2, z_2, y_3, z_3, y_4, z_4, x_5, y_5, z_5, x_6, y_6, z_6$</code>

- (Iwasaki, 1965) found this structure to form metastably between 300° and 600° . Below 300° it transforms into the ordered Au_5Zn_3 structure.

- There is considerable disorder on most of the sites in this system, to wit:
 - Site Au-I (2a) is 60% gold and 40 % zinc,
 - Site Zn-I (2c) is 20% gold and 80 % zinc,
 - Site Zn-II (2b) is 10% gold and 90 % zinc,
 - Site Zn-III (2b) is 40% gold and 60 % zinc,
 - Site Au-II (4c) is 85% gold and 15 % zinc, and
 - Site Au-III (4c) is 100% gold with no zinc.
- We label each site by its dominant component.
- (Iwaskai, 1965) gave the lattice constants in kX units. We used the conversion factor $1 \text{ kX} = 1.00202 \text{ \AA}$. (Wood, 1947).
- The ICSD entry is from (Iwasaki, 1962).

Simple Orthorhombic primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}} \end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= y_1 \mathbf{a}_2 + z_1 \mathbf{a}_3$	$=$	$by_1 \hat{\mathbf{y}} + cz_1 \hat{\mathbf{z}}$	(2a)	Au I
$\mathbf{B}_2$	$= -y_1 \mathbf{a}_2 + (z_1 + \frac{1}{2}) \mathbf{a}_3$	$=$	$-by_1 \hat{\mathbf{y}} + c(z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Au I
$\mathbf{B}_3$	$= y_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	$=$	$by_2 \hat{\mathbf{y}} + cz_2 \hat{\mathbf{z}}$	(2a)	Zn I
$\mathbf{B}_4$	$= -y_2 \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	$=$	$-by_2 \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Zn I
$\mathbf{B}_5$	$= \frac{1}{2} \mathbf{a}_1 + y_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + by_3 \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(2b)	Zn II
$\mathbf{B}_6$	$= \frac{1}{2} \mathbf{a}_1 - y_3 \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - by_3 \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Zn II
$\mathbf{B}_7$	$= \frac{1}{2} \mathbf{a}_1 + y_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + by_4 \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(2b)	Zn III
$\mathbf{B}_8$	$= \frac{1}{2} \mathbf{a}_1 - y_4 \mathbf{a}_2 + (z_4 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - by_4 \hat{\mathbf{y}} + c(z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Zn III
$\mathbf{B}_9$	$= x_5 \mathbf{a}_1 + y_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$ax_5 \hat{\mathbf{x}} + by_5 \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(4c)	Au II
$\mathbf{B}_{10}$	$= -x_5 \mathbf{a}_1 - y_5 \mathbf{a}_2 + (z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$-ax_5 \hat{\mathbf{x}} - by_5 \hat{\mathbf{y}} + c(z_5 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	Au II
$\mathbf{B}_{11}$	$= x_5 \mathbf{a}_1 - y_5 \mathbf{a}_2 + (z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$ax_5 \hat{\mathbf{x}} - by_5 \hat{\mathbf{y}} + c(z_5 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	Au II
$\mathbf{B}_{12}$	$= -x_5 \mathbf{a}_1 + y_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$-ax_5 \hat{\mathbf{x}} + by_5 \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(4c)	Au II
$\mathbf{B}_{13}$	$= x_6 \mathbf{a}_1 + y_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$ax_6 \hat{\mathbf{x}} + by_6 \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(4c)	Au III
$\mathbf{B}_{14}$	$= -x_6 \mathbf{a}_1 - y_6 \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	$=$	$-ax_6 \hat{\mathbf{x}} - by_6 \hat{\mathbf{y}} + c(z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	Au III
$\mathbf{B}_{15}$	$= x_6 \mathbf{a}_1 - y_6 \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	$=$	$ax_6 \hat{\mathbf{x}} - by_6 \hat{\mathbf{y}} + c(z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	Au III
$\mathbf{B}_{16}$	$= -x_6 \mathbf{a}_1 + y_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$-ax_6 \hat{\mathbf{x}} + by_6 \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(4c)	Au III

References

- [1] H. Iwasaki, *The Crystal Structure and the Phase Transition of a Metastable Phase in the Au-37.8% Zn Alloy*, J. Phys. Soc. Jpn. **20**, 2129–2140 (1965), doi:10.1143/JPSJ.20.2129.
- [2] E. A. Wood, *The Conversion Factor for kX Units to Angström Units*, J. Appl. Phys. **18**, 929–930 (1947), doi:10.1063/1.1697570.
- [3] H. Iwasaki, *Study on the Ordered Phases with Long Period in the Gold-Zinc Alloy System II. Structure Analysis of $Au_3Zn[R_1]$, $Au_3Zn[R_2]$ and Au_3+Zn* , J. Phys. Soc. Jpn. **17**, 1620–1633 (1962), doi:10.1143/JPSJ.17.1620.

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