

Al₅C₃N (*E*9₄) Structure: A5B3C_hP18_186_2a3b_2ab_b-001

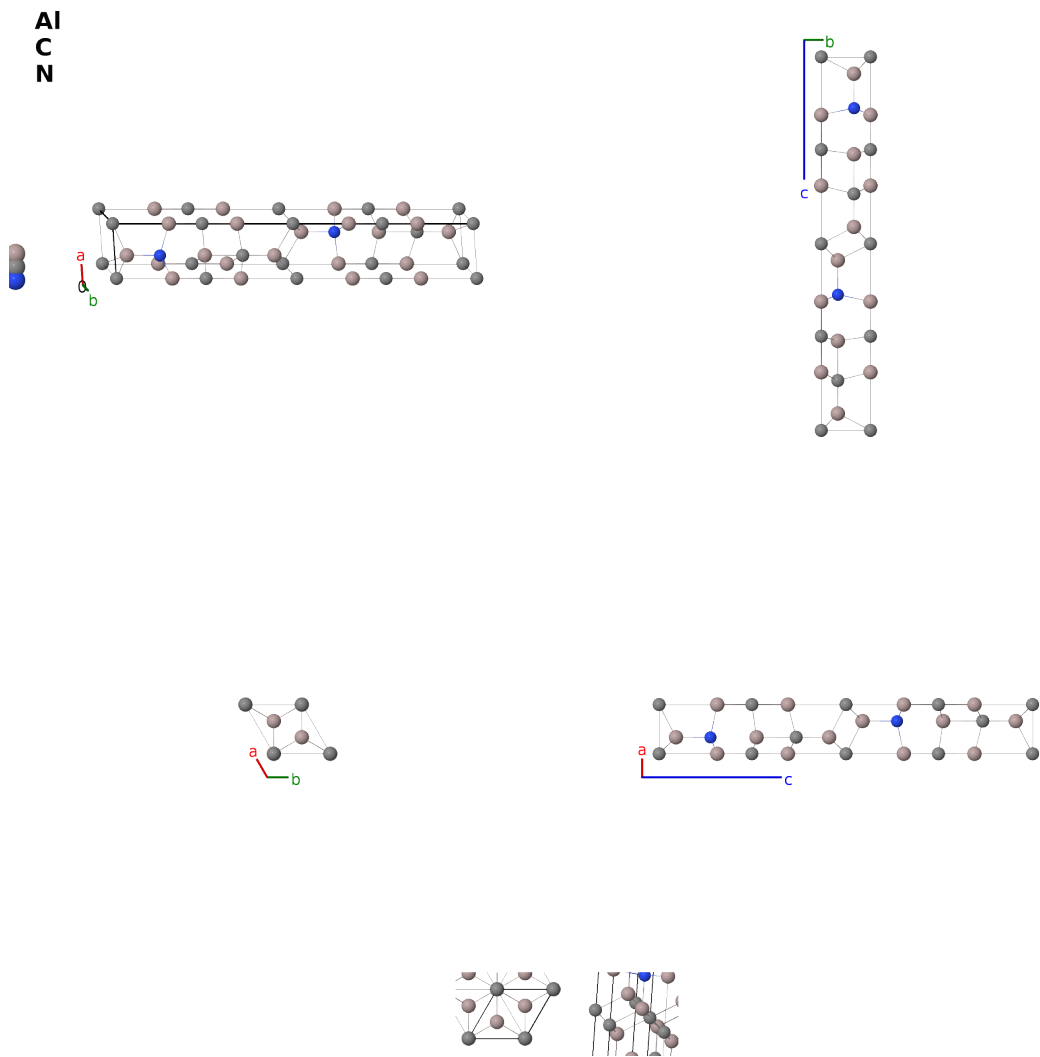
This structure previously used the label(s) **A5B3C_hP18_186_2a3b_2ab_b**. Calls to these addresses will be redirected here.

Cite this page as:

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<https://aflow.org/prototype-encyclopedia/KHLK>

https://aflow.org/prototype-encyclopedia/A5B3C_hP18_186_2a3b_2ab_b-001



Prototype	Al ₅ C ₃ N
AFLOW prototype label	A5B3C_hP18_186_2a3b_2ab_b-001
Strukturbericht designation	<i>E</i> 9 ₄
ICSD	14398

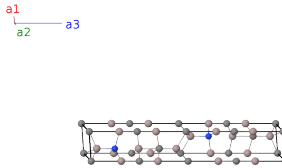
CCDC	1595782
Pearson symbol	hP18
Space group number	186
Space group symbol	$P6_3mc$
AFLOW prototype command	afLOW --proto=A5B3C_hP18_186_2a3b_2ab_b-001 --params= $a, c/a, z_1, z_2, z_3, z_4, z_5, z_6, z_7, z_8, z_9$

Other compounds with this structure

$\text{U}_2\text{Al}_3\text{C}_4$

- Space group $P6_3mc$ #186 allows for an arbitrary placement of the origin of the z -axis. We set this by taking $z_3 = 0$ for the C-I atom.

Hexagonal primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
 \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
 \mathbf{a}_3 &= c \hat{\mathbf{z}}
 \end{aligned}$$


Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= z_1 \mathbf{a}_3$	$=$	$c z_1 \hat{\mathbf{z}}$	(2a)	Al I
$\mathbf{B}_2$	$= (z_1 + \frac{1}{2}) \mathbf{a}_3$	$=$	$c (z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Al I
$\mathbf{B}_3$	$= z_2 \mathbf{a}_3$	$=$	$c z_2 \hat{\mathbf{z}}$	(2a)	Al II
$\mathbf{B}_4$	$= (z_2 + \frac{1}{2}) \mathbf{a}_3$	$=$	$c (z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Al II
$\mathbf{B}_5$	$= z_3 \mathbf{a}_3$	$=$	$c z_3 \hat{\mathbf{z}}$	(2a)	C I
$\mathbf{B}_6$	$= (z_3 + \frac{1}{2}) \mathbf{a}_3$	$=$	$c (z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	C I
$\mathbf{B}_7$	$= z_4 \mathbf{a}_3$	$=$	$c z_4 \hat{\mathbf{z}}$	(2a)	C II
$\mathbf{B}_8$	$= (z_4 + \frac{1}{2}) \mathbf{a}_3$	$=$	$c (z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	C II
$\mathbf{B}_9$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_5 \hat{\mathbf{z}}$	(2b)	Al III
$\mathbf{B}_{10}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c (z_5 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al III
$\mathbf{B}_{11}$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_6 \hat{\mathbf{z}}$	(2b)	Al IV
$\mathbf{B}_{12}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c (z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al IV
$\mathbf{B}_{13}$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_7 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_7 \hat{\mathbf{z}}$	(2b)	Al V
$\mathbf{B}_{14}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_7 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c (z_7 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al V
$\mathbf{B}_{15}$	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_8 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_8 \hat{\mathbf{z}}$	(2b)	C III
$\mathbf{B}_{16}$	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_8 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c (z_8 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	C III

$$\mathbf{B}_{17} = \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_9 \mathbf{a}_3 = \frac{1}{2} a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6} a \hat{\mathbf{y}} + cz_9 \hat{\mathbf{z}} \quad (2b) \quad \text{N I}$$

$$\mathbf{B}_{18} = \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \left(z_9 + \frac{1}{2}\right) \mathbf{a}_3 = \frac{1}{2} a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6} a \hat{\mathbf{y}} + c \left(z_9 + \frac{1}{2}\right) \hat{\mathbf{z}} \quad (2b) \quad \text{N I}$$

References

- [1] G. A. Jeffrey and V. Y. Wu, *The structure of the aluminum carbonitrides. II*, Acta Cryst. **20**, 538–547 (1966), doi:10.1107/S0365110X66001208.

Found in

- [1] P. Villars and K. Cenzual, *Pearson's Crystal Data – Crystal Structure Database for Inorganic Compounds* (2013). ASM International.

First cited in

M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. L. W. Hart, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 1*, Comput. Mater. Sci. **136**, S1-828 (2017). doi: 10.1016/j.commatsci.2017.01.017