

Nd₄Ni₃O₈ Structure:

A4B3C8_tI30_139_2e_ae_cdg-002

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<https://aflow.org/prototype-encyclopedia/W8GW>

https://aflow.org/prototype-encyclopedia/A4B3C8_tI30_139_2e_ae_cdg-002

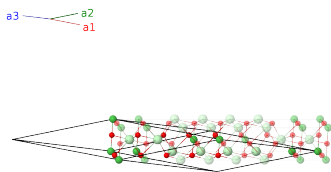
Prototype	Nd ₄ Ni ₃ O ₈
AFLOW prototype label	A4B3C8_tI30_139_2e_ae_cdg-002
ICSD	173372
CCDC	1688346
Pearson symbol	tI30
Space group number	139
Space group symbol	<i>I</i> 4/ <i>mmm</i>
AFLOW prototype command	<code>aflow --proto=A4B3C8_tI30_139_2e_ae_cdg-002 --params=a, c/a, z₄, z₅, z₆, z₇</code>

Other compounds with this structure

La₄Ni₃O₈, Pr₄Ni₃O₈

Body-centered Tetragonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	=	0	=	0	(2a) Ni I
$\mathbf{B}_2$	=	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{y}}$	(4c) O I
$\mathbf{B}_3$	=	$\frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}}$	(4c) O I
$\mathbf{B}_4$	=	$\frac{3}{4}\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(4d) O II

$\mathbf{B}_5$	$=$	$\frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{1}{4} c \hat{\mathbf{z}}$	(4d)	O II
$\mathbf{B}_6$	$=$	$z_4 \mathbf{a}_1 + z_4 \mathbf{a}_2$	$=$	$c z_4 \hat{\mathbf{z}}$	(4e)	Nd I
$\mathbf{B}_7$	$=$	$-z_4 \mathbf{a}_1 - z_4 \mathbf{a}_2$	$=$	$-c z_4 \hat{\mathbf{z}}$	(4e)	Nd I
$\mathbf{B}_8$	$=$	$z_5 \mathbf{a}_1 + z_5 \mathbf{a}_2$	$=$	$c z_5 \hat{\mathbf{z}}$	(4e)	Nd II
$\mathbf{B}_9$	$=$	$-z_5 \mathbf{a}_1 - z_5 \mathbf{a}_2$	$=$	$-c z_5 \hat{\mathbf{z}}$	(4e)	Nd II
$\mathbf{B}_{10}$	$=$	$z_6 \mathbf{a}_1 + z_6 \mathbf{a}_2$	$=$	$c z_6 \hat{\mathbf{z}}$	(4e)	Ni II
$\mathbf{B}_{11}$	$=$	$-z_6 \mathbf{a}_1 - z_6 \mathbf{a}_2$	$=$	$-c z_6 \hat{\mathbf{z}}$	(4e)	Ni II
$\mathbf{B}_{12}$	$=$	$\left(z_7 + \frac{1}{2}\right) \mathbf{a}_1 + z_7 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{y}} + c z_7 \hat{\mathbf{z}}$	(8g)	O III
$\mathbf{B}_{13}$	$=$	$z_7 \mathbf{a}_1 + \left(z_7 + \frac{1}{2}\right) \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + c z_7 \hat{\mathbf{z}}$	(8g)	O III
$\mathbf{B}_{14}$	$=$	$-\left(z_7 - \frac{1}{2}\right) \mathbf{a}_1 - z_7 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{y}} - c z_7 \hat{\mathbf{z}}$	(8g)	O III
$\mathbf{B}_{15}$	$=$	$-z_7 \mathbf{a}_1 - \left(z_7 - \frac{1}{2}\right) \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - c z_7 \hat{\mathbf{z}}$	(8g)	O III

References

- [1] V. V. Poltavets, K. A. Lokshin, M. Croft, T. K. Mandal, T. Egami, and M. Greenblatt, *Crystal Structures of $\text{Ln}_4\text{Ni}_3\text{O}_8$ ($\text{Ln} = \text{La}, \text{Nd}$) Triple Layer T' -type Nickelates*, Inorg. Chem. **46**, 10887–10891 (2007), doi:10.1016/j.intermet.2008.04.018.

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Nd
Ni
O

