

Ni₃Ti (*D*0₂₄) Structure: A3B_hP16_194_gh_ac-001

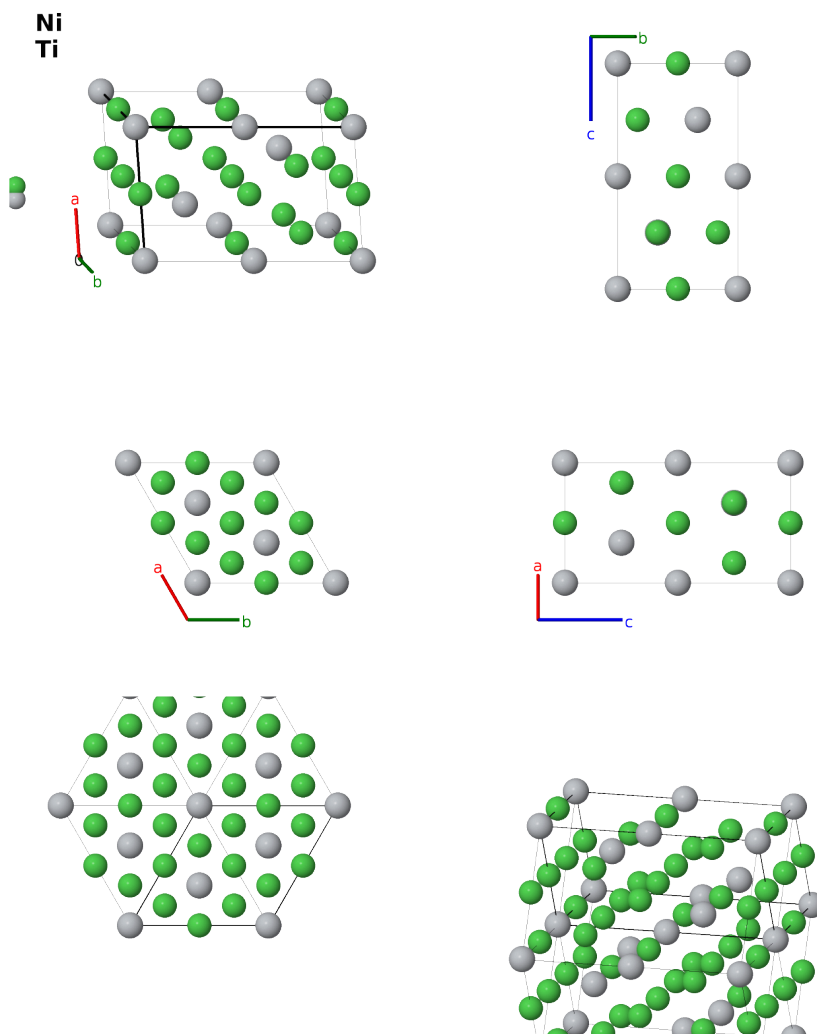
This structure previously used the label(s) **A3B_hP16_194_gh_ac**. Calls to these addresses will be redirected here.

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<https://aflow.org/prototype-encyclopedia/7D2A>

https://aflow.org/prototype-encyclopedia/A3B_hP16_194_gh_ac-001



Prototype

Ni₃Ti

AFLOW prototype label

A3B_hP16_194_gh_ac-001

***Strukturbericht* designation**

*D*0₂₄

ICSD

30216

CCDC	1604588
Pearson symbol	hP16
Space group number	194
Space group symbol	$P6_3/mmc$
AFLOW prototype command	afLOW --proto=A3B_hP16_194_gh_ac-001 --params= $a, c/a, x_4$

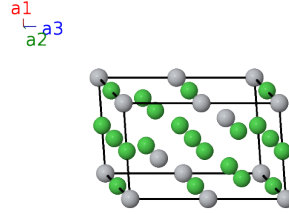
Other compounds with this structure

Al₃Dy, Al₃Ru, Co₃Ti, Ni₃Ti, Pd₃Hf, Pd₃Np, Pd₃Th, Pd₃Ti, Pd₃Zr, Pt₃Hf, Pt₃Zr

- The internal coordinate x_4 was not determined in any reference we could find. We follow (Villars, 2016) and the ICSD and set $x_4 = -1/6$, which places the niobium atoms in line with the Ti atoms in the $z = 1/4$ and $z = 3/4$ planes. This is not required by symmetry, and it is likely that the actual value of x_4 will be close, but not equal to, $-1/6$.

Hexagonal primitive vectors

$$\begin{aligned}
\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
\mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\
\mathbf{a}_3 &= c \hat{\mathbf{z}}
\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$=$	0	$=$	0	(2a) Ti I
$\mathbf{B}_2$	$=$	$\frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(2a) Ti I
$\mathbf{B}_3$	$=$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(2c) Ti II
$\mathbf{B}_4$	$=$	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(2c) Ti II
$\mathbf{B}_5$	$=$	$\frac{1}{2} \mathbf{a}_1$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}}$	(6g) Ni I
$\mathbf{B}_6$	$=$	$\frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}}$	(6g) Ni I
$\mathbf{B}_7$	$=$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{2}a \hat{\mathbf{x}}$	(6g) Ni I
$\mathbf{B}_8$	$=$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6g) Ni I
$\mathbf{B}_9$	$=$	$\frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6g) Ni I
$\mathbf{B}_{10}$	$=$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6g) Ni I
$\mathbf{B}_{11}$	$=$	$x_4 \mathbf{a}_1 + 2x_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h) Ni II
$\mathbf{B}_{12}$	$=$	$-2x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h) Ni II
$\mathbf{B}_{13}$	$=$	$x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h) Ni II
$\mathbf{B}_{14}$	$=$	$-x_4 \mathbf{a}_1 - 2x_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h) Ni II
$\mathbf{B}_{15}$	$=$	$2x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h) Ni II

$$\mathbf{B}_{16} = -x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3 = \sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}} \quad (6h) \quad \text{Ni II}$$

References

- [1] F. Laves and H. J. Wallbaum, *Die Kristallstruktur von Ni_3Ti und Si_2Ti (Zwei neue Typen.)*, Z. Krystallogr. **101**, 78–93 (1939), doi:10.1524/zkri.1939.101.1.78.

Found in

- [1] P. Villars, ed., *PAULING FILE in: Inorganic Solid Phases* (SpringerMaterials (online database), Heidelberg, 2016).

First cited in

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