

Gd₂Cl₃ Structure:

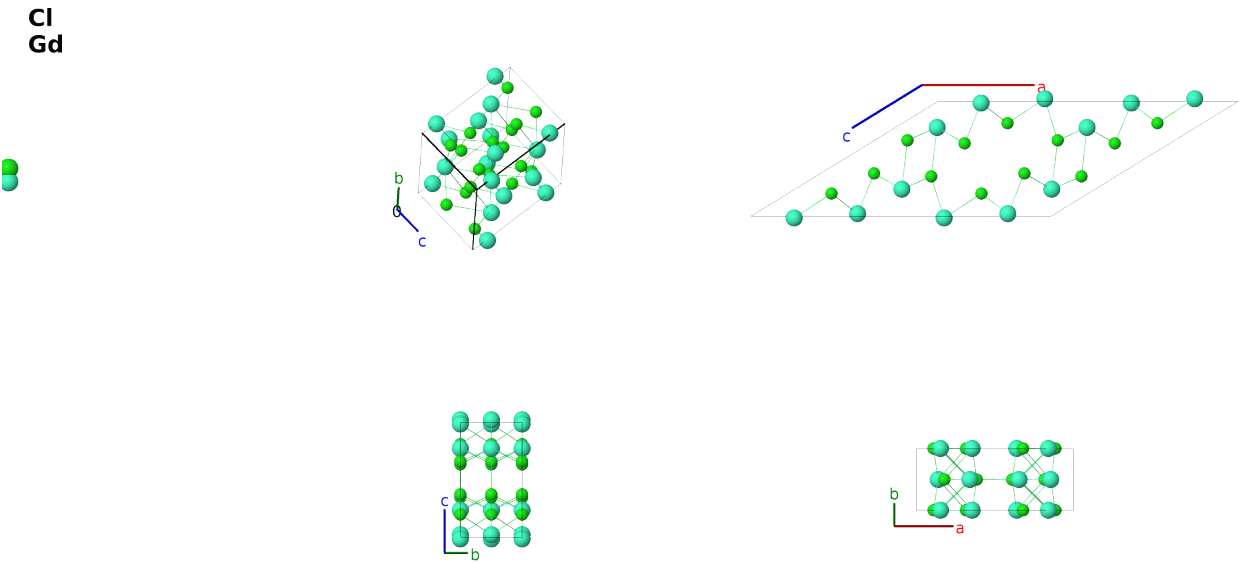
A3B2_mC20_12_3i_2i-002

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<https://aflow.org/prototype-encyclopedia/TUQ7>

https://aflow.org/prototype-encyclopedia/A3B2_mC20_12_3i_2i-002



Prototype	Cl ₃ Gd ₂
AFLOW prototype label	A3B2_mC20_12_3i_2i-002
ICSD	9580
CCDC	1594555
Pearson symbol	mC20
Space group number	12
Space group symbol	$C2/m$
AFLOW prototype command	<code>aflow --proto=A3B2_mC20_12_3i_2i-002</code> <code>--params=a, b/a, c/a, β, $x_1, z_1, x_2, z_2, x_3, z_3, x_4, z_4, x_5, z_5$</code>

Other compounds with this structure

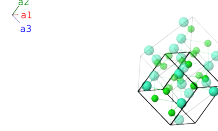
Gd₂Br₃, Tb₂Cl₃, Y₂Cl₃

- There are numerous structures with the AFLOW prototype label A2B3_mC20_12_2i_3i or A3B2_mC20_12_3i_2i. They are generated by the same symmetry operations with different sets of parameters (`--params`) specified in their corresponding CIF files.

- We have identified the following structures as sufficiently different to warrant their own prototypes:
- Prototypes with the label A2B3_mC20_12_2i_3i:
 - β -Ga₂O₃
 - α -As₂Te₃
- Prototypes with the label A3B2_mC20_12_3i_2i:
 - Mo₂As₃
 - Gd₂Cl₃ (this structure)

Base-centered Monoclinic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \cos \beta \hat{\mathbf{x}} + c \sin \beta \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= x_1 \mathbf{a}_1 + x_1 \mathbf{a}_2 + z_1 \mathbf{a}_3$	$=$	$(ax_1 + cz_1 \cos \beta) \hat{\mathbf{x}} + cz_1 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl I
$\mathbf{B}_2$	$= -x_1 \mathbf{a}_1 - x_1 \mathbf{a}_2 - z_1 \mathbf{a}_3$	$=$	$-(ax_1 + cz_1 \cos \beta) \hat{\mathbf{x}} - cz_1 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl I
$\mathbf{B}_3$	$= x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	$=$	$(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} + cz_2 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl II
$\mathbf{B}_4$	$= -x_2 \mathbf{a}_1 - x_2 \mathbf{a}_2 - z_2 \mathbf{a}_3$	$=$	$-(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} - cz_2 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl II
$\mathbf{B}_5$	$= x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} + cz_3 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl III
$\mathbf{B}_6$	$= -x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$-(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} - cz_3 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl III
$\mathbf{B}_7$	$= x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$(ax_4 + cz_4 \cos \beta) \hat{\mathbf{x}} + cz_4 \sin \beta \hat{\mathbf{z}}$	(4i)	Gd I
$\mathbf{B}_8$	$= -x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$-(ax_4 + cz_4 \cos \beta) \hat{\mathbf{x}} - cz_4 \sin \beta \hat{\mathbf{z}}$	(4i)	Gd I
$\mathbf{B}_9$	$= x_5 \mathbf{a}_1 + x_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$(ax_5 + cz_5 \cos \beta) \hat{\mathbf{x}} + cz_5 \sin \beta \hat{\mathbf{z}}$	(4i)	Gd II
$\mathbf{B}_{10}$	$= -x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-(ax_5 + cz_5 \cos \beta) \hat{\mathbf{x}} - cz_5 \sin \beta \hat{\mathbf{z}}$	(4i)	Gd II

References

- [1] A. Simon, N. Holzer, and H. Mattausch, *Metallreiche Verbindungen der Seltenen Erden Gd₂Cl₃, Gd₂Br₃ und Tb₂Cl₃*, Z. Anorganische und Allgemeine Chemie **456**, 207–216 (1979), doi:10.1002/zaac.19794560122.

Found in

- [1] D. Zagorac, H. Müller, S. Ruehl, J. Zagorac, and S. Rehme, *Recent developments in the Inorganic Crystal Structure Database: theoretical crystal structure data and related features*, J. Appl. Crystallogr. **52**, 918–925 (2019), doi:10.1107/S160057671900997X. ICSD-9580 (ICSD release 2023.1).

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