

Mg₂Ni (*C_a*) Structure: A2B_hP18_180_fj_ac-001

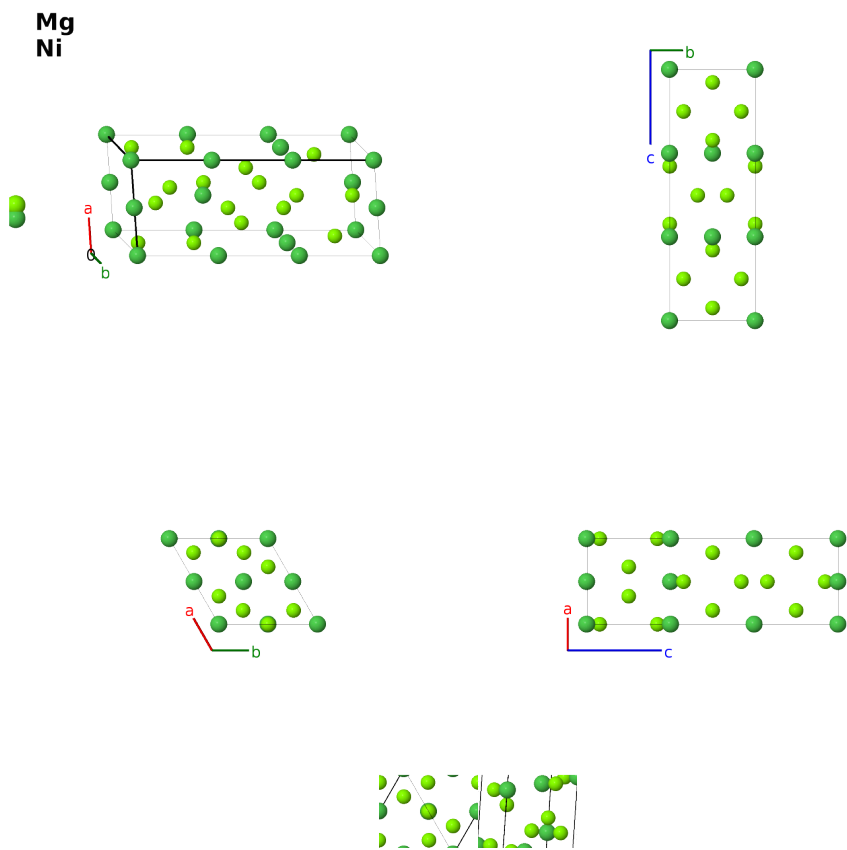
This structure previously used the label(s) **A2B_hP18_180_fi_bd**. Calls to these addresses will be redirected here.

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<https://aflow.org/prototype-encyclopedia/TDY8>

https://aflow.org/prototype-encyclopedia/A2B_hP18_180_fj_ac-001



Prototype	Mg ₂ Ni
AFLOW prototype label	A2B_hP18_180_fj_ac-001
<i>Strukturbericht</i> designation	<i>C_a</i>
ICSD	104912
CCDC	1662813
Pearson symbol	hP18
Space group number	180
Space group symbol	<i>P</i> 6 ₂ 22

AFLOW prototype command `aflow --proto=A2B_hP18_180_fj_ac-001`
 `--params=a, c/a, z3, x4`

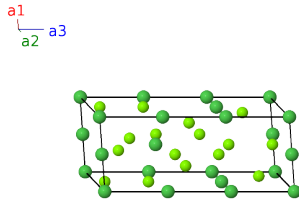
Other compounds with this structure

CuMg₄Ni, MoSn₂

- This structure can also be found in the enantiomorphic space group $P6_422$ #181.
- Our original (Mehl, 2017) rendering of this structure swapped the magnesium z_3 and x_4 coordinates.

Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	0	$=$	0	(3a)	Ni I
$\mathbf{B}_2$	$\frac{2}{3} \mathbf{a}_3$	$=$	$\frac{2}{3}c \hat{\mathbf{z}}$	(3a)	Ni I
$\mathbf{B}_3$	$\frac{1}{3} \mathbf{a}_3$	$=$	$\frac{1}{3}c \hat{\mathbf{z}}$	(3a)	Ni I
$\mathbf{B}_4$	$\frac{1}{2} \mathbf{a}_1$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}}$	(3c)	Ni II
$\mathbf{B}_5$	$\frac{1}{2} \mathbf{a}_2 + \frac{2}{3} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{2}{3}c \hat{\mathbf{z}}$	(3c)	Ni II
$\mathbf{B}_6$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{3} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{3}c \hat{\mathbf{z}}$	(3c)	Ni II
$\mathbf{B}_7$	$\frac{1}{2} \mathbf{a}_1 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(6f)	Mg I
$\mathbf{B}_8$	$\frac{1}{2} \mathbf{a}_2 + (z_3 + \frac{2}{3}) \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{1}{3}c(3z_3 + 2) \hat{\mathbf{z}}$	(6f)	Mg I
$\mathbf{B}_9$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + (z_3 + \frac{1}{3}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + c(z_3 + \frac{1}{3}) \hat{\mathbf{z}}$	(6f)	Mg I
$\mathbf{B}_{10}$	$\frac{1}{2} \mathbf{a}_2 - (z_3 - \frac{2}{3}) \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} - \frac{1}{3}c(3z_3 - 2) \hat{\mathbf{z}}$	(6f)	Mg I
$\mathbf{B}_{11}$	$\frac{1}{2} \mathbf{a}_1 - z_3 \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(6f)	Mg I
$\mathbf{B}_{12}$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 - (z_3 - \frac{1}{3}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - c(z_3 - \frac{1}{3}) \hat{\mathbf{z}}$	(6f)	Mg I
$\mathbf{B}_{13}$	$x_4 \mathbf{a}_1 + 2x_4 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6j)	Mg II
$\mathbf{B}_{14}$	$-2x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{6} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{6}c \hat{\mathbf{z}}$	(6j)	Mg II
$\mathbf{B}_{15}$	$x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{5}{6} \mathbf{a}_3$	$=$	$-\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{5}{6}c \hat{\mathbf{z}}$	(6j)	Mg II
$\mathbf{B}_{16}$	$-x_4 \mathbf{a}_1 - 2x_4 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6j)	Mg II
$\mathbf{B}_{17}$	$2x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{1}{6} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{6}c \hat{\mathbf{z}}$	(6j)	Mg II
$\mathbf{B}_{18}$	$-x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{5}{6} \mathbf{a}_3$	$=$	$\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{5}{6}c \hat{\mathbf{z}}$	(6j)	Mg II

References

- [1] J. Schefer, P. Fischer, W. Hälg, F. Stucki, L. Schlapbach, J. J. Didisheim, K. Yvon, and A. F. Andresen, *New structure results for hydrides and deuterides of the hydrogen storage material Mg_2Ni* , J. Less-Common Met. **74**, 65–73 (1980), doi:10.1016/0022-5088(80)90074-0.
- [2] M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. Hart, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 1*, Comput. Mater. Sci. **136**, S1–S828 (2017), doi:10.1016/j.commatsci.2017.01.017.

Found in

- [1] P. Villars, *Mg_2Ni Crystal Structure* (2016). PAULING FILE in: Inorganic Solid Phases, SpringerMaterials (online database), Springer, Heidelberg (ed.) Springer Materials.

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