

Cuprite (Cu_2O , $C3$) Structure: A2B_cP6_224_b_a-001

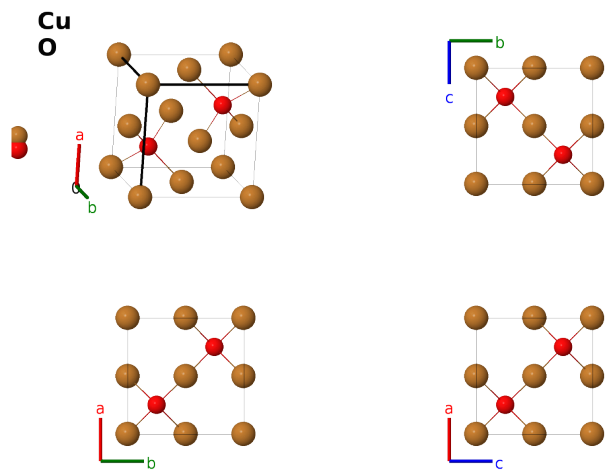
This structure previously used the label(s) **A2B_cP6_224_b_a**. Calls to these addresses will be redirected here.

Cite this page as:

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<https://aflow.org/prototype-encyclopedia/WUJH>

https://aflow.org/prototype-encyclopedia/A2B_cP6_224_b_a-001



Prototype	Cu_2O
AFLOW prototype label	A2B_cP6_224_b_a-001
<i>Strukturbericht</i> designation	$C3$
Mineral name	cuprite
ICSD	52043
CCDC	1616888
Pearson symbol	cP6
Space group number	224
Space group symbol	$Pn\bar{3}m$
AFLOW prototype command	<code>aflow --proto=A2B_cP6_224_b_a-001</code> <code>--params=a</code>

Other compounds with this structure

Ag_2O , Pb_2O

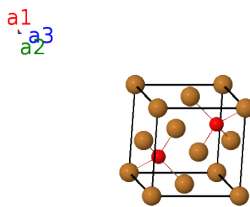
- (Restori, 1986) gives the equilibrium lattice constant of Cu_2O as $a=4.627\text{\AA}$, but gives nearest-neighbor distances which yield a lattice constant of $4.267\text{\AA}$. Since this value agrees with other sources, including those in (Downs, 2003), we use it. The ICSD entry uses $4.2685\text{\AA}$.

Simple Cubic primitive vectors

$$\mathbf{a}_1 = a \hat{\mathbf{x}}$$

$$\mathbf{a}_2 = a \hat{\mathbf{y}}$$

$$\mathbf{a}_3 = a \hat{\mathbf{z}}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{y}} + \frac{1}{4}a \hat{\mathbf{z}}$	(2a)	O I
$\mathbf{B}_2$	$= \frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{3}{4}a \hat{\mathbf{y}} + \frac{3}{4}a \hat{\mathbf{z}}$	(2a)	O I
$\mathbf{B}_3$	$= 0$	$=$	0	(4b)	Cu I
$\mathbf{B}_4$	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}}$	(4b)	Cu I
$\mathbf{B}_5$	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{z}}$	(4b)	Cu I
$\mathbf{B}_6$	$= \frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}a \hat{\mathbf{z}}$	(4b)	Cu I

References

- [1] R. Restori and D. Schwarzenbach, *Charge Density in Cuprite, Cu_2O* , Acta Crystallogr. Sect. B **42**, 201–208 (1986), doi:10.1107/S0108768186098336.
- [2] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).

Found in

- [1] A. Kirfel and K. Eichhorn, *Accurate structure analysis with synchrotron radiation. The electron density in Al_2O_3 and Cu_2O* , Acta Crystallogr. Sect. A **46**, 271–284 (1990), doi:10.1107/S0108767389012596.

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