

Ideal β -Cristobalite (SiO_2 , $C9$) Structure:

A2B_cF24_227_c_a-001

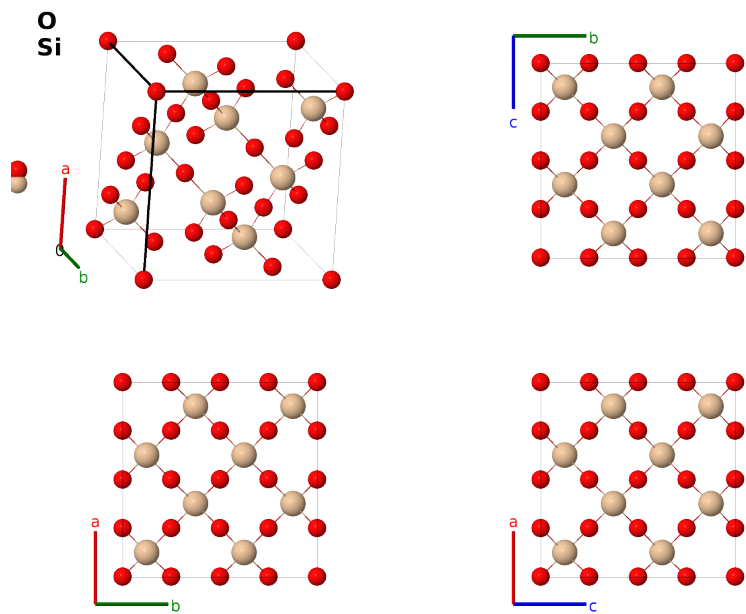
This structure previously used the label(s) **A2B_cF24_227_c_a**. Calls to these addresses will be redirected here.

Cite this page as:

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<https://aflow.org/prototype-encyclopedia/QDSV>

https://aflow.org/prototype-encyclopedia/A2B_cF24_227_c_a-001



Prototype	O_2Si
AFLOW prototype label	A2B_cF24_227_c_a-001
<i>Strukturbericht</i> designation	$C9$
Mineral name	cristobalite
ICSD	35536
CCDC	1607613
Pearson symbol	cF24
Space group number	227
Space group symbol	$Fd\bar{3}m$
AFLOW prototype command	<code>aflow --proto=A2B_cF24_227_c_a-001</code> <code>--params=a</code>

Other compounds with this structure

BeF₂, AlPO₄

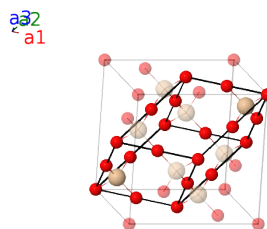
- This is an idealized version of the high-temperature phase of β -cristobalite originally proposed by (Wyckoff, 1925).
 - Below 530K it transforms into the α cristobalite ($C30$) structure (Pluth, 1985).
 - (Peacor, 1973) concludes that the oxygen atoms partially occupy the (96g) positions in the space group $Fd\bar{3}m$ #227. We average those positions to put the oxygen on the (16c) sites and return to the original structure.
 - β -cristobalite is also known as “high” cristobalite.
 - More information about SiO₂ structures can be found in our article on SiO₂ and AlPO₄.
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Face-centered Cubic primitive vectors

$$\mathbf{a}_1 = \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}}$$

$$\mathbf{a}_2 = \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}}$$

$$\mathbf{a}_3 = \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{1}{8}\mathbf{a}_1 + \frac{1}{8}\mathbf{a}_2 + \frac{1}{8}\mathbf{a}_3$	$=$	$\frac{1}{8}a\hat{\mathbf{x}} + \frac{1}{8}a\hat{\mathbf{y}} + \frac{1}{8}a\hat{\mathbf{z}}$	(8a)	Si I
$\mathbf{B}_2$	$= \frac{7}{8}\mathbf{a}_1 + \frac{7}{8}\mathbf{a}_2 + \frac{7}{8}\mathbf{a}_3$	$=$	$\frac{7}{8}a\hat{\mathbf{x}} + \frac{7}{8}a\hat{\mathbf{y}} + \frac{7}{8}a\hat{\mathbf{z}}$	(8a)	Si I
$\mathbf{B}_3$	$= 0$	$=$	0	(16c)	O I
$\mathbf{B}_4$	$= \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{y}}$	(16c)	O I
$\mathbf{B}_5$	$= \frac{1}{2}\mathbf{a}_2$	$=$	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{z}}$	(16c)	O I
$\mathbf{B}_6$	$= \frac{1}{2}\mathbf{a}_1$	$=$	$\frac{1}{4}a\hat{\mathbf{y}} + \frac{1}{4}a\hat{\mathbf{z}}$	(16c)	O I

References

- [1] R. W. G. Wyckoff, *The crystal structure of the high temperature form of cristobalite (SiO₂)*, Am. J. Science **9**, 448–459 (1925).
- [2] D. R. Peacor, *High-temperature single-crystal study of the cristobalite inversion*, Z. Krystallogr. **138**, 274–298 (1973), doi:10.1524/zkri.1973.138.1-4.274.
- [3] J. J. Pluth, J. V. Smith, and J. J. Faber, *Crystal structure of low cristobalite at 10, 293, and 473 K: Variation of framework geometry with temperature*, J. Appl. Phys. **57**, 1045–1049 (1985), doi:10.1063/1.334545.

Found in

- [1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).

First cited in

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