

CaBe₂Ge₂ Structure:

A2BC2_tP10_129_ac_c_bc-001

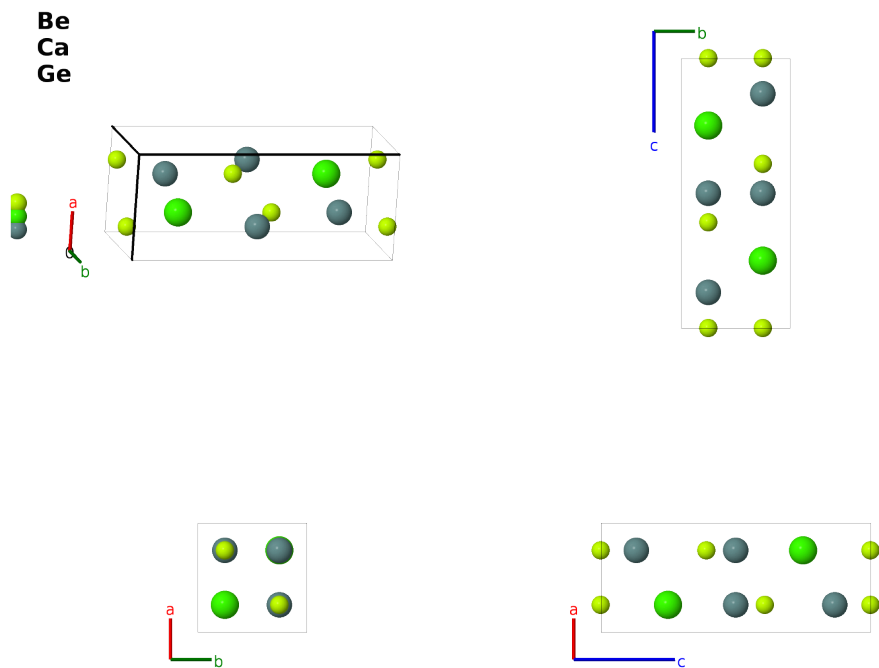
This structure previously used the label(s) **A2BC2_tP10_129_ac_c_bc**. Calls to these addresses will be redirected here.

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<https://aflow.org/prototype-encyclopedia/H0MM>

https://aflow.org/prototype-encyclopedia/A2BC2_tP10_129_ac_c_bc-001



Prototype	Be ₂ CaGe ₂
AFLOW prototype label	A2BC2_tP10_129_ac_c_bc-001
ICSD	25337
CCDC	1601082
Pearson symbol	tP10
Space group number	129
Space group symbol	$P4/nmm$
AFLOW prototype command	<code>aflow --proto=A2BC2_tP10_129_ac_c_bc-001</code> <code>--params=a, c/a, z₃, z₄, z₅</code>

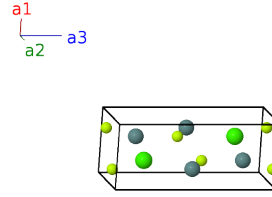
Other compounds with this structure

BaAu₂Sn₂, BaMg₂Pb₂, BaPd₂Sb₂, BaZn₂Sn₂, CeCu₂Sn₂, CeRh₂As₂, CeRh₂P₂, EuAu₂Al₂, EuPd₂Sb₂, EuPt₂Ge₂, HoPt₂Si₂, LaCu₂Sn₂, LaPt₂Bi₂, LaPt₂Ge₂, LaPt₂Si₂, LaRh₂As₂, LaRh₂P₂, LiPd₂Bi₂, NdRh₂As₂, NdRh₂P₂, PrRh₂As₂, PrRh₂P₂, SrAu₂Sn₂, SrCu₂Sn₂, SrPd₂Sb₂, SrPt₂As₂, ThIr₂Si₂, ThPt₂Si₂, UIr₂Si₂, UPt₂Si₂

- This is a ternary form of the $D1_3$ (BaAl₄) structure. The atomic positions are approximately the same as in the conventional cell of BaAl₄, but the distribution of the atoms on those sites and the resulting relaxation leads to a different structure.
- Space group $P4/nmm$ #129 has two settings, but both have the same z -axis origin, so either setting will do here. We chose our standard setting 2.

Simple Tetragonal primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}} \end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{3}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} + \frac{1}{4} a \hat{\mathbf{y}}$	(2a)	Be I
$\mathbf{B}_2$	$= \frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + \frac{3}{4} a \hat{\mathbf{y}}$	(2a)	Be I
$\mathbf{B}_3$	$= \frac{3}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} + \frac{1}{4} a \hat{\mathbf{y}} + \frac{1}{2} c \hat{\mathbf{z}}$	(2b)	Ge I
$\mathbf{B}_4$	$= \frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + \frac{3}{4} a \hat{\mathbf{y}} + \frac{1}{2} c \hat{\mathbf{z}}$	(2b)	Ge I
$\mathbf{B}_5$	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + \frac{1}{4} a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(2c)	Be II
$\mathbf{B}_6$	$= \frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} + \frac{3}{4} a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(2c)	Be II
$\mathbf{B}_7$	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + \frac{1}{4} a \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(2c)	Ca I
$\mathbf{B}_8$	$= \frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} + \frac{3}{4} a \hat{\mathbf{y}} - cz_4 \hat{\mathbf{z}}$	(2c)	Ca I
$\mathbf{B}_9$	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + \frac{1}{4} a \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(2c)	Ge II
$\mathbf{B}_{10}$	$= \frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} + \frac{3}{4} a \hat{\mathbf{y}} - cz_5 \hat{\mathbf{z}}$	(2c)	Ge II

References

- [1] B. Eisenmann, N. May, W. Müller, and H. Schäfer, *Eine neue strukturelle Variante des BaAl₄-Typs: Der CaBe₂Ge₂-Typ*, Z. Naturforsch. B **27** (1972), doi:10.1515/znb-1972-1008. 1155-1157.

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