

α -AgI (*B23*) Structure: A21B_cI44_229_bdh_a-001

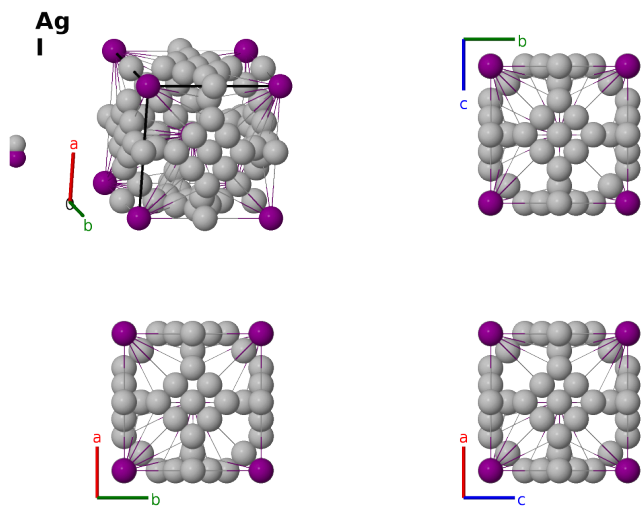
This structure previously used the label(s) **A21B_cI44_229_bdh_a**. Calls to these addresses will be redirected here.

Cite this page as:

- H. Eckert, S. Divilov, M. J. Mehl, D. Hicks, A. C. Zettel, M. Esters, X. Campilongo, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*, Comput. Mater. Sci. **240**, 112988 (2024), doi: 10.1016/j.commatsci.2024.112988.
- D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/prototype-encyclopedia/QV5K>

https://aflow.org/prototype-encyclopedia/A21B_cI44_229_bdh_a-001



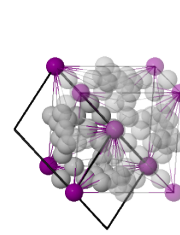
Prototype	AgI
AFLOW prototype label	A21B_cI44_229_bdh_a-001
<i>Strukturbericht</i> designation	<i>B23</i>
ICSD	33626
CCDC	1606383
Pearson symbol	cI44
Space group number	229
Space group symbol	$Im\bar{3}m$
AFLOW prototype command	<code>aflow --proto=A21B_cI44_229_bdh_a-001</code> <code>--params=a, y_4</code>

Other compounds with this structure

Ag₂S, Ag₂Se

- Under ambient conditions, silver iodide exists as a mixture of β -AgI, which has the wurtzite (*B4*) structure, and γ -AgI, which has the zincblende (*B3*) structure (Hull, 2004). Above 420K AgI transforms to this superionic α phase. The iodine atom sits at the (2a) site of the bcc lattice of space group $Im\bar{3}m$ #229, while the silver atom is randomly distributed on one of the (6b), (12d), and (24h) Wyckoff sites in each unit cell. On average, then, each of the 21 Ag sites listed below is occupied only 4.762% of the time in any given primitive cell. This easy transport between sites drives the superionic behavior of α -AgI. Ag₂S and Ag₂Se have higher concentrations of silver on each site.
- There is no ICSD entry for (Strock, 1934), so we use that from (Rahlf, 1936) for AgS. The two are otherwise identical except for the lattice constant ($a = 5.034\text{\AA}$ for Strock, $4.88\text{\AA}$ for Rahlf. We still chose AgI over AgS for the prototype label because *Strukturbericht* designated the former as the prototype.

Body-centered Cubic primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= -\frac{1}{2}a\hat{x} + \frac{1}{2}a\hat{y} + \frac{1}{2}a\hat{z} \\
 \mathbf{a}_2 &= \frac{1}{2}a\hat{x} - \frac{1}{2}a\hat{y} + \frac{1}{2}a\hat{z} \\
 \mathbf{a}_3 &= \frac{1}{2}a\hat{x} + \frac{1}{2}a\hat{y} - \frac{1}{2}a\hat{z}
 \end{aligned}$$


Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$=$	0	$=$	0	I I
$\mathbf{B}_2$	$=$	$\frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{x}$	(6b) Ag I
$\mathbf{B}_3$	$=$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{y}$	(6b) Ag I
$\mathbf{B}_4$	$=$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2$	$=$	$\frac{1}{2}a\hat{z}$	(6b) Ag I
$\mathbf{B}_5$	$=$	$\frac{1}{2}\mathbf{a}_1 + \frac{3}{4}\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{x} + \frac{1}{2}a\hat{z}$	(12d) Ag II
$\mathbf{B}_6$	$=$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{x} + \frac{1}{2}a\hat{y}$	(12d) Ag II
$\mathbf{B}_7$	$=$	$\frac{1}{4}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{x} + \frac{1}{4}a\hat{y}$	(12d) Ag II
$\mathbf{B}_8$	$=$	$\frac{3}{4}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{y} + \frac{1}{2}a\hat{z}$	(12d) Ag II
$\mathbf{B}_9$	$=$	$\frac{3}{4}\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{y} + \frac{1}{4}a\hat{z}$	(12d) Ag II
$\mathbf{B}_{10}$	$=$	$\frac{1}{4}\mathbf{a}_1 + \frac{3}{4}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{x} + \frac{1}{4}a\hat{z}$	(12d) Ag II
$\mathbf{B}_{11}$	$=$	$2y_4\mathbf{a}_1 + y_4\mathbf{a}_2 + y_4\mathbf{a}_3$	$=$	$ay_4\hat{y} + ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{12}$	$=$	$y_4\mathbf{a}_2 - y_4\mathbf{a}_3$	$=$	$-ay_4\hat{y} + ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{13}$	$=$	$-y_4\mathbf{a}_2 + y_4\mathbf{a}_3$	$=$	$ay_4\hat{y} - ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{14}$	$=$	$-2y_4\mathbf{a}_1 - y_4\mathbf{a}_2 - y_4\mathbf{a}_3$	$=$	$-ay_4\hat{y} - ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{15}$	$=$	$y_4\mathbf{a}_1 + 2y_4\mathbf{a}_2 + y_4\mathbf{a}_3$	$=$	$ay_4\hat{x} + ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{16}$	$=$	$-y_4\mathbf{a}_1 + y_4\mathbf{a}_3$	$=$	$ay_4\hat{x} - ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{17}$	$=$	$y_4\mathbf{a}_1 - y_4\mathbf{a}_3$	$=$	$-ay_4\hat{x} + ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{18}$	$=$	$-y_4\mathbf{a}_1 - 2y_4\mathbf{a}_2 - y_4\mathbf{a}_3$	$=$	$-ay_4\hat{x} - ay_4\hat{z}$	(24h) Ag III
$\mathbf{B}_{19}$	$=$	$y_4\mathbf{a}_1 + y_4\mathbf{a}_2 + 2y_4\mathbf{a}_3$	$=$	$ay_4\hat{x} + ay_4\hat{y}$	(24h) Ag III
$\mathbf{B}_{20}$	$=$	$y_4\mathbf{a}_1 - y_4\mathbf{a}_2$	$=$	$-ay_4\hat{x} + ay_4\hat{y}$	(24h) Ag III
$\mathbf{B}_{21}$	$=$	$-y_4\mathbf{a}_1 + y_4\mathbf{a}_2$	$=$	$ay_4\hat{x} - ay_4\hat{y}$	(24h) Ag III

$$\mathbf{B}_{22} = -y_4 \mathbf{a}_1 - y_4 \mathbf{a}_2 - 2y_4 \mathbf{a}_3 = -ay_4 \hat{\mathbf{x}} - ay_4 \hat{\mathbf{y}} \quad (24h) \quad \text{Ag III}$$

References

- [1] L. W. Strock, *Kristallstruktur des Hochtemperatur-Jodsilbers α -AgI*, Z. Physik. Chem. B **25**, 441–459 (1934), doi:10.1515/zpch-1934-2535.
- [2] P. Rahlfs, *Über die kubischen Hochtemperaturmodifikationen der Sulfide, Selenide und Telluride des Silbers und des einwertigen Kupfers*, Z. Physik. Chem. B **31**, 157–194 (1936), doi:10.1515/zpch-1936-3114.
- [3] S. Hull, *Superionics: crystal structures and conduction processes*, Rep. Prog. Phys. **67**, 1233–1314 (2004), doi:10.1088/0034-4885/67/7/R05.

Found in

- [1] S. Hoshino, *Crystal Structure and Phase Transition of Some Metallic Halides IV On the Anomalous Structure of α -AgI*, J. Phys. Soc. Japan **12**, 315–326 (1957), doi:10.1143/JPSJ.12.315.

First cited in

D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.