

UB₁₂ (*D*2_{*f*}) Structure: A12B_cF52_225_h_b-001

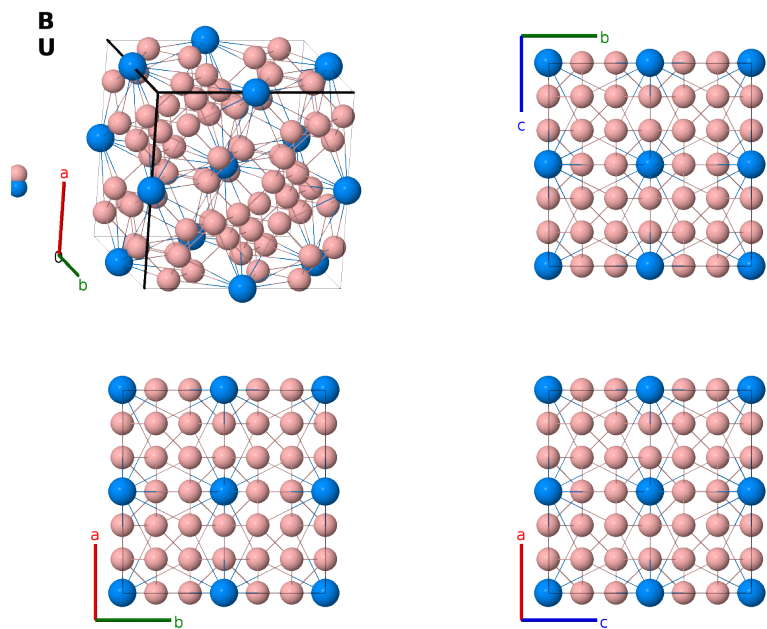
This structure previously used the label(s) **A12B_cF52.225.i.a.** Calls to these addresses will be redirected here.

Cite this page as:

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<https://aflow.org/prototype-encyclopedia/JTKK>

https://aflow.org/prototype-encyclopedia/A12B_cF52_225_h_b-001



Prototype	B ₁₂ U
AFLOW prototype label	A12B_cF52_225_h_b-001
<i>Strukturbericht</i> designation	<i>D</i> 2 _{<i>f</i>}
ICSD	24705
CCDC	1600718
Pearson symbol	cF52
Space group number	225
Space group symbol	<i>Fm</i> $\bar{3}$ <i>m</i>
AFLOW prototype command	<code>aflow --proto=A12B_cF52_225_h_b-001</code> <code>--params=<i>a</i>, <i>y</i>₂</code>

Other compounds with this structure

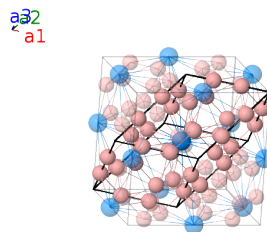
DyB₁₂, ErB₁₂, LuB₁₂, ThB₁₂, TmB₁₂, YB₁₂, YbB₁₂, ZrB₁₂, (Th_{0.93}Zr_{0.07})B₁₂

Face-centered Cubic primitive vectors

$$\mathbf{a}_1 = \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}a \hat{\mathbf{z}}$$

$$\mathbf{a}_2 = \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{z}}$$

$$\mathbf{a}_3 = \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}a \hat{\mathbf{z}}$	(4b)	U I
$\mathbf{B}_2$	$= 2y_2 \mathbf{a}_1$	$=$	$ay_2 \hat{\mathbf{y}} + ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_3$	$= 2y_2 \mathbf{a}_2 - 2y_2 \mathbf{a}_3$	$=$	$-ay_2 \hat{\mathbf{y}} + ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_4$	$= -2y_2 \mathbf{a}_2 + 2y_2 \mathbf{a}_3$	$=$	$ay_2 \hat{\mathbf{y}} - ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_5$	$= -2y_2 \mathbf{a}_1$	$=$	$-ay_2 \hat{\mathbf{y}} - ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_6$	$= 2y_2 \mathbf{a}_2$	$=$	$ay_2 \hat{\mathbf{x}} + ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_7$	$= -2y_2 \mathbf{a}_1 + 2y_2 \mathbf{a}_3$	$=$	$ay_2 \hat{\mathbf{x}} - ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_8$	$= 2y_2 \mathbf{a}_1 - 2y_2 \mathbf{a}_3$	$=$	$-ay_2 \hat{\mathbf{x}} + ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_9$	$= -2y_2 \mathbf{a}_2$	$=$	$-ay_2 \hat{\mathbf{x}} - ay_2 \hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_{10}$	$= 2y_2 \mathbf{a}_3$	$=$	$ay_2 \hat{\mathbf{x}} + ay_2 \hat{\mathbf{y}}$	(48h)	B I
$\mathbf{B}_{11}$	$= 2y_2 \mathbf{a}_1 - 2y_2 \mathbf{a}_2$	$=$	$-ay_2 \hat{\mathbf{x}} + ay_2 \hat{\mathbf{y}}$	(48h)	B I
$\mathbf{B}_{12}$	$= -2y_2 \mathbf{a}_1 + 2y_2 \mathbf{a}_2$	$=$	$ay_2 \hat{\mathbf{x}} - ay_2 \hat{\mathbf{y}}$	(48h)	B I
$\mathbf{B}_{13}$	$= -2y_2 \mathbf{a}_3$	$=$	$-ay_2 \hat{\mathbf{x}} - ay_2 \hat{\mathbf{y}}$	(48h)	B I

References

- [1] P. Blum and F. Bertaut, *Contribution à l'Étude des Borures à Teneur Élevée en Bore*, Acta Cryst. **7**, 81–86 (1954), doi:10.1107/S0365110X54000151.

Found in

- [1] W. B. Pearson, *The Crystal Chemistry and Physics of Metals and Alloys* (Wiley Interscience, New York, London, Sydney, Toronto, 1972).

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